

=> fil reg
FILE 'REGISTRY' ENTERED AT 09:35:41 ON 02 MAR 2010
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STRUCTURE FILE UPDATES: 1 MAR 2010 HIGHEST RN 1207596-35-7
DICTIONARY FILE UPDATES: 1 MAR 2010 HIGHEST RN 1207596-35-7

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TSCA INFORMATION NOW CURRENT THROUGH June 26, 2009.

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<http://www.cas.org/support/stngen/stndoc/properties.html>

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=> d que
L2      20 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON (1333-74-0/B1 OR
          15318-08-8/B1 OR 19553-62-9/B1 OR 20791-15-5/B1 OR
          310888-77-8/B1 OR 310888-80-3/B1 OR 310888-81-4/B1 OR
          310888-82-5/B1 OR 310888-85-8/B1 OR 310888-87-0/B1 OR
          7358-26-1/B1 OR 7440-37-1/B1 OR 7440-59-7/B1 OR 75-24-1/B1
          OR 7727-37-9/B1 OR 870126-56-0/B1 OR 870126-57-1/B1 OR
          870126-58-2/B1 OR 870126-59-3/B1 OR 97-93-8/B1)
L3      1 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON 8-HYDROXYQUINOLIN
          ATE/CN
L4      1 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON 8-HYDROXYQUINOLIN
          E/CN
L5      685 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON 148-24-3/CRN
L6      7 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON L5 AND (AL OR GA
          OR ZN)/ELS
L7      16 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON L2 AND M/ELS
L8      663827 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON (AL OR GA OR
          ZN)/ELS
L9      1222525 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON L8 AND CCS/CI
L10     541302 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON L8 NOT L9
L11     541302 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON L10 OR L10
L12     300000 SEA FILE=REGISTRY RAN=(173351-91-2) SPE=ON ABB=ON PLU=ON
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L13     241302 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON L11 NOT L12
L14     23243 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L7
L15     143187 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L9
L16     140299 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L12
L17     2254991 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L13
L18     33 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON L8 AND HYDROXYQUI
          NOL?
L19     9 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON 16582-16-4/CRN
L20     1 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON L19 AND (AL OR
          GA OR ZN)/ELS
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L21 30 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L3
 L22 10121 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L4
 L23 1836 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L5
 L24 9 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L19
 L25 6 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L6
 L26 9679 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L18
 L27 1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L20
 L28 11763 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L14 OR L15 OR
 L16 OR L17) AND (L21 OR L22 OR L23 OR L24 OR L25 OR L26 OR
 L27)
 L29 QUE SPE=ON ABB=ON PLU=ON LUM!N? OR ELECTROLUM!N? OR O
 RGANOLUM!N? OR (ELECTRO OR ORGANO OR ORG#) (2A)LUM!N? OR L
 IGHT? (2A) (EMIT? OR EMISSION?) OR EL OR E(W)L OR L(W)E(W)D
 OR OLED OR LED
 L30 8689 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L28 AND L29
 L31 2171 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L30 AND PROC/RL
 L32 246823 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON "VAPOR DEPOSITION
 PROCESS"+PFT,NT/CT
 L33 251 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L31 AND L32
 L34 12 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L33 AND (L21 OR
 L22 OR L23)
 L35 414 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L30 AND (L21 OR
 L22 OR L23)
 L36 89 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L35 AND L31
 L37 12 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L36 AND CPS/RL
 L38 11 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L36 AND (VAPOR
 DEPOSIT? OR VAPOUR DEPOSIT?)
 L39 23 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L34 OR (L37 OR
 L38)
 L40 QUE SPE=ON ABB=ON PLU=ON REACTOR# OR (REACTION#) (2A)
 (VESSEL# OR CHAMBER# OR TANK# OR SYSTEM# OR SPACE# OR CO
 MPARTMENT# OR RECEPTACLE# OR PORTION# OR PORT# OR ASSEMBL
 Y# OR SUB# (W)ASSEMBLY#)
 L41 1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L36 AND L40
 L42 7 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L31 AND L40
 L43 29 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L39 OR L41 OR L42

 L44 20 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L43 AND (1840-2006
)/PRY,AY,PY
 L45 258 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L30 AND (L21 OR
 L22)
 L48 71 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L45 AND (FEED? OR
 DELIVER? OR SUPPLY? OR DISTRIBUT? OR TRANSPORT?)
 L49 58 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L48 AND OPTIC?/SC,
 SX
 L50 49 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L49 AND (1840-2006
)/PRY,AY,PY
 L53 QUE SPE=ON ABB=ON PLU=ON FILM? OR THINFILM? OR LAYER?
 OR OVERLAY? OR OVERLAID? OR LAMIN? OR LAMEL? OR (MULTILA
 YER?) OR SHEET? OR LEAF? OR FOIL? OR COAT? OR TOPCOAT? OR
 OVERCOAT? OR VENEER? OR SHEATH? OR COVER? OR ENVELOP? OR
 ENCAS? OR ENWRAP? OR OVERSPREAD?
 L55 49 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L7 AND (L21 OR
 L22 OR L23)
 L56 10 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L55 AND L29 AND
 L53
 L57 21 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L55 AND L29
 L58 21 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L56 OR L57
 L59 18 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L58 AND (1840-2006
)/PRY,AY,PY

L60 38 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L44 OR L59
 L61 9 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L60 AND L50
 L62 38 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L60 OR L61)

=> fil hcap
 FILE 'HCAPLUS' ENTERED AT 09:35:46 ON 02 MAR 2010
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FILE COVERS 1907 - 2 Mar 2010 VOL 152 ISS 10
 FILE LAST UPDATED: 1 Mar 2010 (20100301/ED)
 REVISED CLASS FIELDS (/NCL) LAST RELOADED: Dec 2009
 USPTO MANUAL OF CLASSIFICATIONS THESAURUS ISSUE DATE: Dec 2009

HCAplus now includes complete International Patent Classification (IPC) reclassification data for the third quarter of 2009.

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This file contains CAS Registry Numbers for easy and accurate substance identification.

=> d 162 1-38 ibib ed abs hitstr hitind

L62 ANSWER 1 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2008:379801 HCAPLUS Full-text
 DOCUMENT NUMBER: 148:415641
 TITLE: Transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element
 INVENTOR(S): Tateishi, Tomomi
 PATENT ASSIGNEE(S): Fujifilm Corporation, Japan
 SOURCE: U.S. Pat. Appl. Publ., 25 pp.
 CODEN: USXXCO
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 20080075921	A1	20080327	US 2007-902031 ---<--	20070918
JP 2008084701	A	20080410	JP 2006-263437	20060927

PRIORITY APPLN. INFO.:

JP 2006-263437

A 20060927

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ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

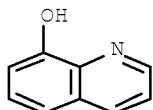
ED Entered STN: 28 Mar 2008

AB The present invention provides a transfer material with a strong adhesiveness for an electronic device that includes a transfer support and, provided on the support in this order, an insulating layer or a partition wall material layer, and a layer containing an organic low-mol.-weight compound having charge transportability; a method of forming an insulating layer and a partition wall of an electronic device using the transfer material; and a light-emitting element.

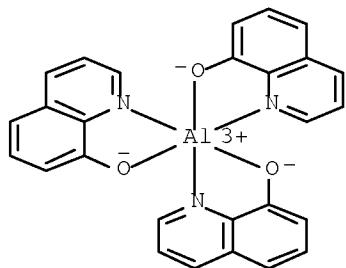
IT 148-24-3D, 8-Quinolinol, derivs. 2085-33-8, Alq3
 (transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element)

RN 148-24-3 HCPLUS

CN 8-Quinolinol (CA INDEX NAME)



RN 2085-33-8 HCPLUS

CN Aluminum, tris(8-quinolinolato- κ N1, κ O8)- (CA INDEX NAME)

INCL 428141000; 156230000; 428172000

CC 76-3 (Electric Phenomena)

Section cross-reference(s): 48, 73

ST transfer electronic device insulating layer partition wall LED fabrication

IT Cluster ions

(beams; transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element)

IT Anhydrides

(dianhydrides; transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element)

IT Vapor deposition process

- (ion plating; transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element)
- IT Materials
 (organic; transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element)
- IT Chemical vapor deposition
 (photochem.; transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element)
- IT Polymerization
 Vapor deposition process
 (plasma; transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element)
- IT Acrylic polymers
 (polysiloxane-, US-3700; transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element)
- IT Conducting polymers
 (polythiophenes; transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element)
- IT Chemical vapor deposition
 Coating process
 Dielectric films
 Electroluminescent devices
 Holders
 Molecular beam epitaxy
 Reactive sputtering
 Release films
 Semiconductor device fabrication
 Sputtering
 Transfers
 (transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element)
- IT Metallophthalocyanines
 Polyanilines
 Polyesters
 Polyphenyls
 (transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element)
- IT Vapor deposition process
 (vacuum; transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element)
- IT 67-63-0, Isopropyl alcohol, processes 78-93-3, Methyl ethyl ketone, processes 108-88-3, Toluene, processes 58328-31-7 60676-86-0, Vitreous silica 475644-38-3, Optool DSX 757974-86-0, TFR-H
 (transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element)
- IT 50926-11-9, Indium tin oxide 128770-43-4, HP 320 (polyester)
 (transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element)
- IT 90-44-8D, Anthrone, derivs. 92-52-4D, Biphenyl, quinone derivs.

147-14-8, Copper phthalocyanine 148-24-3D, 8-Quinolinol, derivs. 151-51-9D, Carbodiimide, derivs. 273-53-0D, Benzoxazole, metal complexes 288-42-6D, Oxazole, derivs. 288-88-0D, 1H-1,2,4-Triazole, derivs. 486-25-9D, Fluorenone, derivs. 574-93-6D, Phthalocyanine, derivs. 2085-33-8, Alq3 4425-82-5D, Fluorenylidenemethane, derivs. 7789-24-4, Lithium fluoride, processes 11120-54-0D, Oxadiazole, derivs. 12597-68-1, Stainless steel, processes 14990-02-4D, derivs. 25038-59-9, Lumirror T-60, processes 60475-00-5D, Thiopyran, derivs. 70359-39-6D, derivs. 95270-88-5D, Polyfluorene, derivs. 96638-49-2D, Polyphenylenevinylene, derivs. 123847-85-8 380234-99-1, ZPN 1100 693794-98-8
 (transfer material for electronic device, method of forming insulating layer and partition wall of electronic device, and light-emitting element)

OS.CITING REF COUNT: 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1 CITINGS)

L62 ANSWER 2 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2007:1109833 HCPLUS Full-text
 DOCUMENT NUMBER: 147:416094
 TITLE: Production method of organic electroluminescent device and electronic apparatus
 INVENTOR(S): Takashima, Takeshi; Terao, Koichi; Shinohara, Takashi
 PATENT ASSIGNEE(S): Seiko Epson Corp., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 63pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2007257898	A	20071004	JP 2006-77832 -----<--	20060320
KR 2007095194	A	20070928	KR 2007-24559 -----<--	20070313
US 20070231467	A1	20071004	US 2007-688128 -----<--	20070319
PRIORITY APPLN. INFO.:			JP 2006-77832 -----<--	A 20060320

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

ED Entered STN: 04 Oct 2007

AB The invention relates to a production method of an organic electroluminescent device that comprises ≥ 2 electroluminescent layers having dissimilar luminescent colors each other and fabricated between a pair of electrodes, comprising the steps of: forming a 1st color electroluminescent layer on a 1st electrode, using an electroluminescent polymerizable compound and a carrier transporting polymerizable compound; irradiating the 1st color electroluminescent layer to cure the polymerizable compds.; and removing the 1st color electroluminescent layer that is not irradiated.

IT 97-93-8, Triethyl aluminum, reactions 148-24-3,
 8-Quinolinol, reactions

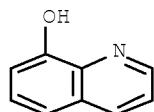
(organic electroluminescent device and electronic apparatus)

RN 97-93-8 HCPLUS

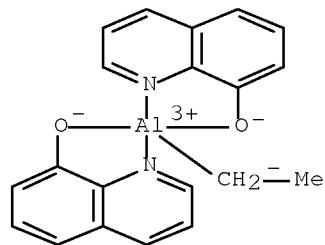
CN Aluminum, triethyl- (CA INDEX NAME)



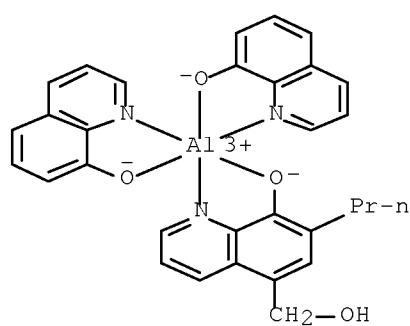
RN 148-24-3 HCAPLUS
 CN 8-Quinolinol (CA INDEX NAME)



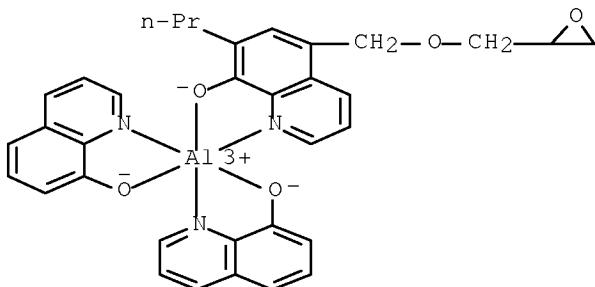
IT 950598-70-6P 950598-71-7P 950598-72-8P
 950598-73-9P
 (organic electroluminescent device and electronic apparatus)
 RN 950598-70-6 HCAPLUS
 CN Aluminum, ethylbis(8-quinolinolato-κN1,κO8)- (CA INDEX NAME)



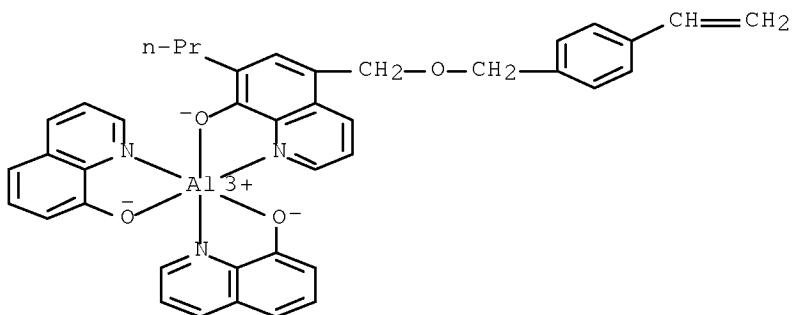
RN 950598-71-7 HCAPLUS
 CN Aluminum, [8-(hydroxy-κO)-7-propyl-5-quinolinemethanolato-κN1]bis(8-quinolinolato-κN1,κO8)- (CA INDEX NAME)



RN 950598-72-8 HCAPLUS
 CN Aluminum, [5-[(2-oxiranylmethoxy)methyl]-7-propyl-8-quinolinolato-
 κN1,κO8]bis(8-quinolinolato-κN1,κO8)- (CA
 INDEX NAME)



RN 950598-73-9 HCAPLUS
 CN Aluminum, [5-[(4-ethenylphenyl)methoxy]methyl]-7-propyl-8-
 quinolinolato-κN1,κO8]bis(8-quinolinolato-
 κN1,κO8)- (CA INDEX NAME)



CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other
 Related Properties)
 Section cross-reference(s): 25, 27, 74
 ST org electroluminescent device electronic app
 IT Electroluminescent devices
 (displays; organic electroluminescent device and electronic
 apparatus)
 IT Luminescent screens
 (electroluminescent; organic
 electroluminescent device and electronic apparatus)
 IT Coating process
 Electric apparatus
 Electroluminescent devices
 (organic electroluminescent device and electronic apparatus)
 IT 64-17-5, Ethanol, reactions 97-93-8, Triethyl aluminum,
 reactions 98-73-7 100-39-0 100-51-6, Benzylalcohol, reactions
 106-38-7 106-89-8, Epichlorohydrin, reactions 108-24-7 108-44-1,

m-Toluidine, reactions 121-43-7, Boric acid trimethyl ester
 137-07-5 148-24-3, 8-Quinolinol, reactions 302-01-2,
 Hydrazine, reactions 507-16-4, Thionyl bromide 612-62-4,
 2-Chloroquinoline 873-75-6, 4-Bromobenzylalcohol 1592-20-7,
 4-Vinylbenzylchloride 2746-25-0, 4-Methoxybenzylbromide 3001-15-8,
 4,4'-Diiodobiphenyl 3006-96-0 3047-32-3 4316-58-9 5029-67-4,
 2-Iodopyridine 5798-75-4 7719-09-7, Thionyl chloride 25574-11-2
 30525-89-4, Paraformaldehyde 58327-60-9, 7-Propyl-8-quinolinol
 59016-93-2 76283-09-5, 4-Bromo-2-fluorobenzylbromide 91251-45-5
 951123-25-4

(organic electroluminescent device and electronic apparatus)

IT 537-92-8P 1484-13-5P, N-Vinylcarbazole 1710-98-1P 14996-61-3P
 49743-87-5P 69135-05-3P 96411-61-9P 421553-47-1P 433332-03-7P
 890844-70-9P 950590-42-8P 950590-43-9P 950590-44-0P
 950590-45-1P 950590-46-2DP, reaction products with epichlorohydrin
 or 4-vinylbenzylchloride 950590-47-3P 950590-49-5P 950590-50-8P
 950590-51-9P 950590-52-0P 950590-53-1P 950590-54-2P
 950590-57-5P 950590-58-6P 950590-59-7P 950590-61-1P
 950590-62-2P 950598-61-5P 950598-62-6P 950598-63-7P
 950598-64-8P 950598-65-9P 950598-66-0P 950598-67-1P
 950598-68-2P 950598-69-3P 950598-70-6P
 950598-71-7P 950598-72-8P 950598-73-9P
 951123-28-7P 951123-29-8P 951123-30-1P 951123-31-2P

(organic electroluminescent device and electronic apparatus)

OS.CITING REF COUNT: 2 THERE ARE 2 CAPLUS RECORDS THAT CITE THIS
 RECORD (2 CITINGS)

L62 ANSWER 3 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2007:1056999 HCAPLUS Full-text
 DOCUMENT NUMBER: 147:395308
 TITLE: Method for manufacturing light
 emitting element, light
 emitting device, and electronic apparatus
 INVENTOR(S): Takashima, Takeshi; Terao, Koichi; Shinohara,
 Takashi
 PATENT ASSIGNEE(S): Seiko Epson Corporation, Japan
 SOURCE: U.S. Pat. Appl. Publ., 37pp.
 CODEN: USXXCO
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 20070218190	A1	20070920	US 2007-688126 --->	20070319
JP 2007257897	A	20071004	JP 2006-77831 --->	20060320
KR 2007095196	A	20070928	KR 2007-24577 --->	20070313
PRIORITY APPLN. INFO.:			JP 2006-77831 --->	A 20060320

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

ED Entered STN: 20 Sep 2007

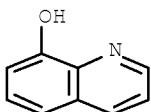
AB A method for manufacturing a light-emitting element, including: forming a film
 on one side of a first electrode; obtaining a light-emitting layer in the film
 by polymerizing a first compound as well as at least one of a second compound
 and a third compound; and installing a second electrode on a side opposite
 from the first electrode in the light-emitting layer; wherein: the first

compound is provided with an emissive light-emitting moiety and a first polymerizable group; the second compound is provided with a hole-transporting hole transport moiety and a second polymerizable group; and the third compound is provided with an electron-transporting electron transport moiety and a third polymerizable group.

IT 97-93-8, Triethylaluminum, reactions 148-24-3,
 8-Quinolinol, reactions
 (method for manufacturing light-emitting element,
 light-emitting device, and electronic apparatus)
 RN 97-93-8 HCAPLUS
 CN Aluminum, triethyl- (CA INDEX NAME)



RN 148-24-3 HCAPLUS
 CN 8-Quinolinol (CA INDEX NAME)



INCL 427066000; 427064000
 CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
 ST manuf light emitting element device electronic app
 IT Electroluminescent devices
 (method for manufacturing light-emitting element,
 light-emitting device, and electronic apparatus)
 IT 97-93-8, Triethylaluminum, reactions 98-73-7,
 p-t-Butylbenzoic acid 100-39-0, Benzyl bromide 106-89-8,
 Epichlorohydrin, reactions 137-07-5, 2-Aminothiophenol
 148-24-3, 8-Quinolinol, reactions 302-01-2, Hydrazine,
 reactions 507-16-4, Thionyl bromide 612-62-4, 2-Chloroquinoline
 873-75-6, 4-Bromobenzyl alcohol 1592-20-7, 4-Vinylbenzyl chloride
 2156-04-9, 4-Vinylphenylboronic acid 2746-25-0, 4-Methoxybenzyl
 bromide 3006-96-0, 4-Hydroxymethylbenzoic acid 3047-32-3,
 3-Ethyl-3-hydroxymethyloxetane 5029-67-4, 2-Iodopyridine
 5798-75-4, 4-Bromobenzoic acid ethyl ester 7719-09-7, Thionyl
 chloride 14996-61-3, Iridium trichloride hydrate 30030-25-2
 30525-89-4, Paraformaldehyde 58327-60-9, 7-Propyl-8-quinolinol
 59016-93-2, 4-(Hydroxymethyl)phenylboronic acid 76283-09-5,
 4-Bromo-2-fluorobenzyl bromide
 (method for manufacturing light-emitting element,
 light-emitting device, and electronic apparatus)
 IT 1710-98-1P 49743-87-5P 69135-05-3P 96411-61-9P 421553-47-1P
 890844-70-9P 950590-42-8P 950590-43-9P 950590-44-0P
 950590-45-1P 950590-46-2P 950590-47-3P 950590-49-5P
 950590-50-8P 950590-51-9P 950590-52-0P 950590-53-1P
 950590-57-5P 950590-59-7P 950598-61-5P 950598-62-6P

950598-64-8P 950598-69-3P 950598-70-6P 950598-71-7P

950598-73-9P

(method for manufacturing light-emitting element,
light-emitting device, and electronic apparatus)

IT 950590-48-4P 950590-54-2P 950590-55-3P 950590-56-4P
950590-58-6P 950590-61-1P 950590-62-2P 950598-63-7P
950598-65-9P 950598-66-0P 950598-67-1P 950598-68-2P
950598-72-8P

(method for manufacturing light-emitting element,
light-emitting device, and electronic apparatus)

IT 7429-90-5, Aluminum, uses 7440-70-2, Calcium, uses
(method for manufacturing light-emitting element,
light-emitting device, and electronic apparatus)

OS.CITING REF COUNT: 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS
RECORD (1 CITINGS)

L62 ANSWER 4 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN

ACCESSION NUMBER: 2006:1178076 HCAPLUS Full-text

DOCUMENT NUMBER: 146:89819

TITLE: Enhanced performance in organic light-emitting diodes with copolymers containing both tris(8-hydroxyquinoline) aluminum and 8-hydroxyquinoline lithium groups

AUTHOR(S): Du, N. Y.; Tian, R. Y.; Peng, J. B.; Mei, Q. B.; Lu, M. G.

CORPORATE SOURCE: Key Laboratory of Polymer Materials for Electronics, Guangdong Guangzhou Institute of Chemistry, Chinese Academy of Sciences, Guangzhou, 510650, Peop. Rep. China

SOURCE: Journal of Applied Polymer Science (2006), 102(5), 4404-4410

CODEN: JAPNAB; ISSN: 0021-8995

PUBLISHER: John Wiley & Sons, Inc.

DOCUMENT TYPE: Journal

LANGUAGE: English

ED Entered STN: 09 Nov 2006

AB The authors synthesized novel copolymers containing both tris(8-hydroxyquinoline) Al (Alq₃) and 8-hydroxyquinoline Li (Liq) groups as emitting layers for use in conventional two-layer organic light-emitting diodes. The network structure and thermal stability of these materials is described. The optical and electroluminescent properties of the copolymers were also studied. The performance optimization of the devices with the copolymers through the variation of the ratio of Alq₃ to Liq is described. A mechanism responsible for the improved electron injection is put forward.

IT 97-93-8, Triethylaluminum, reactions 148-24-3,

8-Hydroxyquinoline, reactions

(enhanced performance in organic light-emitting diodes with copolymers containing both tris(8-hydroxyquinoline) aluminum and 8-hydroxyquinoline lithium groups)

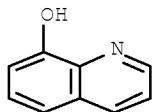
RN 97-93-8 HCAPLUS

CN Aluminum, triethyl- (CA INDEX NAME)



RN 148-24-3 HCAPLUS

CN 8-Quinolinol (CA INDEX NAME)



CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
 ST org LED copolymer tris hydroxyquinoline aluminum lithium
 IT Electroluminescence
 Electroluminescent devices
 Glass substrates
 Luminescence
 Optical absorption
 UV and visible spectra
 (enhanced performance in organic light-emitting
 diodes with copolymers containing both tris(8-hydroxyquinoline)
 aluminum and 8-hydroxyquinoline lithium groups)
 IT 50926-11-9, Indium tin oxide 126213-51-2, PEDOT
 (enhanced performance in organic light-emitting
 diodes with copolymers containing both tris(8-hydroxyquinoline)
 aluminum and 8-hydroxyquinoline lithium groups)
 IT 97-93-8, Triethylaluminum, reactions 127-09-3, Sodium
 acetate 148-24-3, 8-Hydroxyquinoline, reactions
 150-76-5, p-Methoxyphenol 868-77-9 1310-65-2, Lithium hydroxide
 (enhanced performance in organic light-emitting
 diodes with copolymers containing both tris(8-hydroxyquinoline)
 aluminum and 8-hydroxyquinoline lithium groups)
 IT 847506-46-1P 851593-62-9P
 (enhanced performance in organic light-emitting
 diodes with copolymers containing both tris(8-hydroxyquinoline)
 aluminum and 8-hydroxyquinoline lithium groups)
 IT 847506-46-1DP, derivs. 851593-62-9DP, derivs.
 (enhanced performance in organic light-emitting
 diodes with copolymers containing both tris(8-hydroxyquinoline)
 aluminum and 8-hydroxyquinoline lithium groups)
 IT 7429-90-5, Aluminum, uses
 (enhanced performance in organic light-emitting
 diodes with copolymers containing both tris(8-hydroxyquinoline)
 aluminum and 8-hydroxyquinoline lithium groups)
 OS.CITING REF COUNT: 2 THERE ARE 2 CAPLUS RECORDS THAT CITE THIS
 RECORD (2 CITINGS)
 REFERENCE COUNT: 22 THERE ARE 22 CITED REFERENCES AVAILABLE FOR
 THIS RECORD. ALL CITATIONS AVAILABLE IN THE
 RE FORMAT

L62 ANSWER 5 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2006:717587 HCPLUS Full-text
 DOCUMENT NUMBER: 145:176961
 TITLE: Preparation of aluminum
 tris(8-hydroxyquinoline)/silicon oxide complex
 luminescent material with improved
 chemical stability and luminous
 intensity
 INVENTOR(S): Zeng, Hongyu; Shi, Jianlin

PATENT ASSIGNEE(S): Shanghai Institute of Ceramics, Chinese Academy of Sciences, Peop. Rep. China
 SOURCE: Faming Zhuanli Shengqing Gongkai Shuomingshu, 9 pp.
 CODEN: CNXXEV
 DOCUMENT TYPE: Patent
 LANGUAGE: Chinese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
CN 1803971	A	20060719	CN 2005-10112410 <-- CN 2005-10112410 <--	20051230

PRIORITY APPLN. INFO.: CN 2005-10112410 20051230
 <--

ED Entered STN: 25 Jul 2006

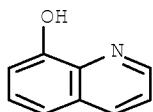
AB The title luminescent material for organic light emitting diode (OLED) device is prepared by: (1) grafting organic reactive group onto 8-hydroxyquinoline (HQ) for functional modification, (2) dissolving in ethanol, and adding silane coupling agent to obtain a silanized complex (SiHQ), (3) mixing with DMF, adding tetra-Et orthosilicate (TEOS), deionized water, and Al³⁺ solution, stirring, and drying to obtain the final product, wherein the molar ratio of SiHQ to TEOS is 1:(5-60). The process is simple and can be carried out in solution. The obtained material has an emission spectrum within blue light range and has advantages of high luminescent intensity, high chemical stability to water and oxygen, long service life, and low cost.

IT 148-24-3, 8-Hydroxyquinoline, processes 22537-23-1

, Aluminum 3+, processes
(preparation of aluminum tris(8-hydroxyquinoline)/silicon oxide complex luminescent material with improved chemical stability and luminous intensity)

RN 148-24-3 HCAPLUS

CN 8-Quinolinol (CA INDEX NAME)



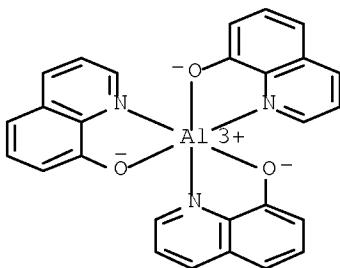
RN 22537-23-1 HCAPLUS
 CN Aluminum, ion (Al³⁺) (CA INDEX NAME)

Al³⁺

CC 73-5 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
 ST aluminum hydroxyquinoline silicon oxide complex luminescent material prepn
 IT Coupling agents
 Luminescent substances
 (preparation of aluminum tris(8-hydroxyquinoline)/silicon oxide complex

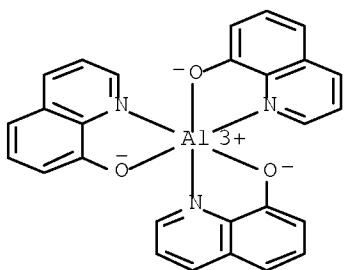
- luminescent material with improved chemical stability and
luminous intensity)
- IT Silanes
 (preparation of aluminum tris(8-hydroxyquinoline)/silicon oxide complex
luminescent material with improved chemical stability and
luminous intensity)
- IT 78-10-4, Tetraethyl orthosilicate 148-24-3,
 8-Hydroxyquinoline, processes 22537-23-1, Aluminum 3+,
 processes
 (preparation of aluminum tris(8-hydroxyquinoline)/silicon oxide complex
luminescent material with improved chemical stability and
luminous intensity)

L62 ANSWER 6 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2006:674276 HCPLUS [Full-text](#)
 DOCUMENT NUMBER: 145:301732
 TITLE: Design and synthesis of Alq₃-functionalized SBA-15
 mesoporous material
 AUTHOR(S): Wang, Hongsu; Huang, Jiahui; Wu, Shujie; Xu, Chen;
 Xing, Lihong; Xu, Ling; Kan, Qiubin
 CORPORATE SOURCE: Department of Chemistry, Jilin University,
 Changchun, 130023, Peop. Rep. China
 SOURCE: Materials Letters (2006), 60(21-22),
 2662-2665
 CODEN: MLETDJ; ISSN: 0167-577X
 PUBLISHER: Elsevier B.V.
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 ED Entered STN: 13 Jul 2006
 AB A luminescent organic mol. aluminum tris(8-hydroxyquinoline) (Alq₃) has been
 successfully introduced in the pores of amino-functionalized mesoporous SBA-15
 (APS-SBA-15). An obvious blue-shifted photoluminescence (PL) of Alq₃ was
 observed In the pores of APS-SBA-15, the Alq₃ organic mols. exhibited
 efficient and intense PL as monomers and the emission intensities increased
 with increasing Alq₃ concentration
 IT 2085-33-8DP, silane-base
 (design and synthesis of Alq₃-functionalized SBA-15 mesoporous
 material)
 RN 2085-33-8 HCPLUS
 CN Aluminum, tris(8-quinolinolato-κN1,κO8)- (CA INDEX NAME)



- IT 2085-33-8
 (design and synthesis of Alq₃-functionalized SBA-15 mesoporous
 material)
 RN 2085-33-8 HCPLUS

CN Aluminum, tris(8-quinolinolato- κ N1, κ O8)- (CA INDEX NAME)



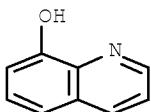
IT 97-93-8, Triethyl aluminum, reactions 148-24-3,
8-Hydroxyquinoline, reactions
(design and synthesis of Alq3-functionalized SBA-15 mesoporous material)

RN 97-93-8 HCPLUS

CN Aluminum, triethyl- (CA INDEX NAME)



RN 148-24-3 HCPLUS
CN 8-Quinolinol (CA INDEX NAME)



CC 73-5 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
Section cross-reference(s): 78

ST Alq3 functionalized SBA mesoporous material luminescence pore distribution synthesis

IT Luminescent substances
(design and synthesis of Alq3-functionalized SBA-15 mesoporous material)

IT Luminescence
Pore size
Pore size distribution
Surface structure
X-ray diffraction
(of Alq3-functionalized SBA-15 mesoporous material)

IT 2085-33-8DP, silane-base
(design and synthesis of Alq3-functionalized SBA-15 mesoporous material)

IT 2085~33~8
 (design and synthesis of Alq3-functionalized SBA-15 mesoporous material)

IT 97~93~8, Triethyl aluminum, reactions 148~24~3,
 8-Hydroxyquinoline, reactions 919~30~2, 3-Aminopropyltriethoxysilane
 2598~30~3, 5-Formyl-8-hydroxyquinoline
 (design and synthesis of Alq3-functionalized SBA-15 mesoporous material)

OS.CITING REF COUNT: 7 THERE ARE 7 CAPLUS RECORDS THAT CITE THIS RECORD (7 CITINGS)

REFERENCE COUNT: 21 THERE ARE 21 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L62 ANSWER 7 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2006:452762 HCAPLUS Full-text
 DOCUMENT NUMBER: 145:104019
 TITLE: End-capped polystyrene with 8-hydroxyquinoline group by ATRP method
 AUTHOR(S): Liu, Cheng-Mei; Qiu, Jin-Jun; Bao, Rui; Xu, Yan;
 Cheng, Xiao-Ju; Hu, Fen
 CORPORATE SOURCE: Department of Chemistry, Huazhong University of Science and Technology, Wuhan, 430074, Peop. Rep. China
 SOURCE: Polymer (2006), 47(9), 2962-2969
 CODEN: POLMAG; ISSN: 0032-3861

PUBLISHER: Elsevier Ltd.

DOCUMENT TYPE: Journal

LANGUAGE: English

ED Entered STN: 16 May 2006

AB 8-Hydroxyquinoline end-capped polystyrene was prepared through atom transfer free radical polymerization (ATRP) with 8-(5-chloromethyl) quinolyl acetate as initiator. The results indicated that this polymerization is a first order reaction with respect to monomer conversion. The mol. weight increased linearly with monomer consumption and very narrow distribution of mol. weight was obtained (polydispersity index less than 1.2). The FT-IR and NMR results show that the 8-hydroxyquinoline group was chemical bonded to the polymer end and there is nearly one 8-hydroxyquinoline group in per polymer chain. All those data show that polymerization of styrene at such conditions displayed living characters. The polymer with 8-hydroxyquinoline end group reacted with triethylaluminum to form polymeric light-emitting complex and single layer LED was prepared by common spin-coating method. The peak wavelength of LED based on synthesized polymeric complex was around 570 nm.

IT 97~93~8DF, Triethylaluminum, derivative with 8-hydroxyquinoline derivative capped polystyrene
 (ATRP kinetics of styrene catalyzed with 8-hydroxyquinoline derivative)

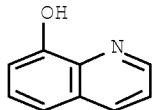
RN 97-93-8 HCAPLUS

CN Aluminum, triethyl- (CA INDEX NAME)



IT 148~24~3, 8-Hydroxyquinoline, reactions
 (in preparation of initiator containing 8-hydroxyquinoline group for ATRP
 of
 styrene)

RN 148-24-3 HCAPLUS
 CN 8-Quinolinol (CA INDEX NAME)



CC 35-4 (Chemistry of Synthetic High Polymers)
 Section cross-reference(s): 73
 ST light emitting coating hydroxyquinoline
 capped polystyrene ATRP
 IT Electroluminescent devices
 (LED prepared with end-capped polystyrene containing
 8-hydroxyquinoline group)
 IT 97-93-80P, Triethylaluminum, derivative with 8-hydroxyquinoline
 derivative capped polystyrene
 (ATRP kinetics of styrene catalyzed with 8-hydroxyquinoline derivative)
 IT 50-00-0, Formaldehyde, reactions 75-36-5, Acetyl chloride
 148-24-3, 8-Hydroxyquinoline, reactions 7647-01-0, Hydrogen
 chloride, reactions
 (in preparation of initiator containing 8-hydroxyquinoline group for ATRP
 of
 styrene)
 OS.CITING REF COUNT: 5 THERE ARE 5 CAPLUS RECORDS THAT CITE THIS
 RECORD (5 CITINGS)
 REFERENCE COUNT: 28 THERE ARE 28 CITED REFERENCES AVAILABLE FOR
 THIS RECORD. ALL CITATIONS AVAILABLE IN THE
 RE FORMAT

L62 ANSWER 8 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2006:138274 HCAPLUS Full-text
 DOCUMENT NUMBER: 144:282934
 TITLE: Method for preparing aluminum 8-hydroxyquinoline
 complex as blue-light organic
 electroluminescent material
 INVENTOR(S): Xu, Bingshe; Wang, Hua; Hao, Yuying; Zhou, Hefeng;
 Li, Jie; Ruan, Limin
 PATENT ASSIGNEE(S): Taiyuan University of Technology, Peop. Rep. China
 SOURCE: Faming Zhanli Shengqing Gongkai Shuomingshu, 19
 Pp.
 CODEN: CNXXXEV
 DOCUMENT TYPE: Patent
 LANGUAGE: Chinese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
CN 1597669	A	20050323	CN 2004-10012430	20040719
CN 1283628	C	20061108	<--	
PRIORITY APPLN. INFO.:			CN 2004-10012430	20040719

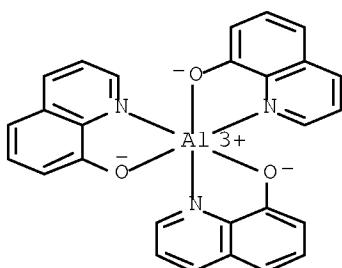
ED Entered STN: 15 Feb 2006

AB The title organic electroluminescent material is prepared from 8-hydroxyquinoline crystal, anhydrous aluminum acetate, anhydrous methanol, acetone, N,N'-dimethyl formamide and pyridine by carrying out reaction under nitrogen protection, washing, filtering, drying, vacuum heating, maintaining the temperature, cooling, purifying by recrystn., and analyzing to obtain the final product. This method has advantages of simple process, plentiful raw materials, high product quality, low manufacturing cost, etc.

IT 2085-33-8P, Alq₃
 (method for preparing aluminum 8-hydroxyquinoline complex as blue-light organic electroluminescent material)

RN 2085-33-8 HCPLUS

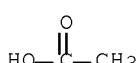
CN Aluminum, tris(8-quinolinolato-κN1,κO8)- (CA INDEX NAME)



IT 139-12-8, Aluminum acetate 148-24-3,
 8-Hydroxyquinoline, reactions
 (method for preparing aluminum 8-hydroxyquinoline complex as blue-light organic electroluminescent material)

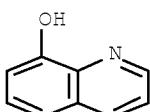
RN 139-12-8 HCPLUS

CN Acetic acid, aluminum salt (3:1) (CA INDEX NAME)



●1/3 Al

RN 148-24-3 HCPLUS
 CN 8-Quinolinol (CA INDEX NAME)



IC ICM C07D215-24
 ICS C07F005-06

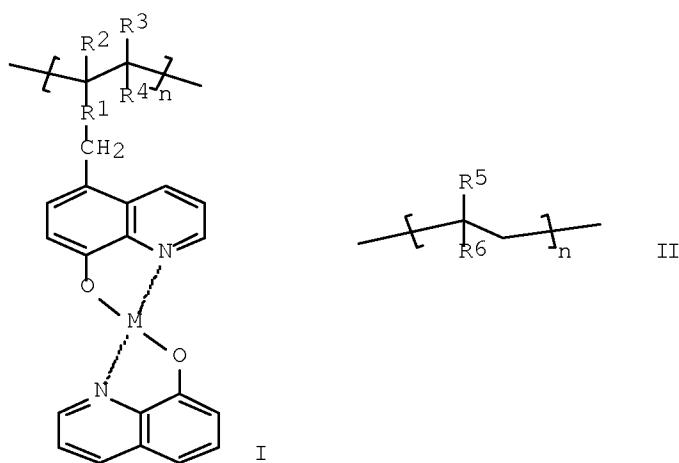
CC 73-5 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
 Section cross-reference(s): 27
 ST aluminum hydroxyquinoline complex blue org **electroluminescent**
 material prepn
 IT Luminescent substances
 (electroluminescent; method for preparing aluminum
 8-hydroxyquinoline complex as blue-light organic
 electroluminescent material)
 IT Recrystallization
 (method for preparing aluminum 8-hydroxyquinoline complex as
 blue-light organic **electroluminescent** material)
 IT 2085~33~8P, Alq3
 (method for preparing aluminum 8-hydroxyquinoline complex as
 blue-light organic **electroluminescent** material)
 IT 64-19-7, Acetic acid, formation (nonpreparative)
 (method for preparing aluminum 8-hydroxyquinoline complex as
 blue-light organic **electroluminescent** material)
 IT 67-56-1, Methanol, uses 67-64-1, Acetone, uses 68-12-2, uses
 110-86-1, Pyridine, uses 7727-37-9, Nitrogen, uses
 (method for preparing aluminum 8-hydroxyquinoline complex as
 blue-light organic **electroluminescent** material)
 IT 139-12-8, Aluminum acetate 148-24-3,
 8-Hydroxyquinoline, reactions
 (method for preparing aluminum 8-hydroxyquinoline complex as
 blue-light organic **electroluminescent** material)

L62 ANSWER 9 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2006:136159 HCPLUS Full-text
 DOCUMENT NUMBER: 144:192599
 TITLE: Preparation of monomers and copolymers containing
 8-hydroxyquinoline metal complexes
 INVENTOR(S): Du, Naiying; Lu, Mangeng; Mei, Qunbo
 PATENT ASSIGNEE(S): Guangzhou Institute of Chemistry, Chinese Academy
 of Sciences, Peop. Rep. China
 SOURCE: Faming Zhanli Shenqing Gongkai Shuomingshu, 13
 pp.
 CODEN: CNXXEV
 DOCUMENT TYPE: Patent
 LANGUAGE: Chinese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
CN 1594382	A	20050316	CN 2004-10027706 -->	20040618
CN 1240733	C	20060208	CN 2004-10027706 -->	20040618

PRIORITY APPLN. INFO.:

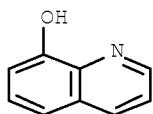
ED Entered STN: 14 Feb 2006
 GI



- AB A linear polymer with a mol. weight of 20000-100000, contains structural unit (I) and (II) at a molar ratio of 1:0.5-200, wherein M is a metal, R1 is C2-18 alkyl, ester, ether or amino, R2, R3, R4, R5 and R6 are selected from H and C1-18 alkyl. This copolymer is prepared by free-radical polymerization of 1 part 8-hydroxyquinoline metal complex containing vinyl monomer with one polymerizable functional group and 1-1000 parts vinyl monomer (e.g., allyl alc., vinylphenyl chloride, acryloyl chloride, styrene, C3-18 unsatd. acid, C3-18 unsatd. ester) in 5-500 parts polar organic solvent (e.g., anhydrous ethanol, THF, toluene, xylene, chloroform, benzene) in the presence of 0.01-0.5 parts free-radical initiator (e.g., BPO, AIBN) at 20-200° for 1-20 h, precipitating with methanol. The polymer may be used as luminescent material in organic electroluminescence flat panel display device and electron transport material. Thus, a luminescent polymer is prepared by copolymerg. 60 g 2-hydroxyethyl methacrylate with 10.0 g 8-hydroxyquinoline aluminum complex of Al³⁺, 2-[(8-hydroxy-5-quinolinyl)methoxy]ethyl methacrylate and 8-hydroxyquinoline.
- IT 97-93-8, Triethyl aluminum, reactions 148-24-3,
8-Hydroxyquinoline, reactions
(preparation of monomers and copolymers containing 8-hydroxyquinoline metal complexes)
- RN 97-93-8 HCPLUS
- CN Aluminum, triethyl- (CA INDEX NAME)



- RN 148-24-3 HCPLUS
- CN 8-Quinolinol (CA INDEX NAME)



IC ICM C08F226-06
 ICS C09K011-06; C08F002-06; C08F004-34
 CC 35-2 (Chemistry of Synthetic High Polymers)
 Section cross-reference(s): 76
 ST hydroxyquinoline aluminum complex prepn; luminescent
 hydroxyquinoline complex hydroxyethyl methacrylate copolymer
 IT Electroluminescent devices
 (displays; preparation of monomers and copolymers containing
 8-hydroxyquinoline metal complexes)
 IT Luminescent screens
 (electroluminescent; preparation of monomers and copolymers
 containing 8-hydroxyquinoline metal complexes)
 IT 75-01-4, Vinyl chloride, reactions 97-93-8, Triethyl
 aluminum, reactions 140-24-3, 8-Hydroxyquinoline,
 reactions 555-31-7, Isopropanol Aluminumsalt 557-34-6, Zinc
 acetate 868-77-9, 2-Hydroxyethyl methacrylate 920-46-7,
 Methacrylic chloride 1310-65-2, Lithium hydroxide 1577-22-6,
 5-Hexenoic acid 1592-20-7, 4-Vinylbenzylchloride 4053-44-5,
 8-Hydroxy-5-Hydroxymethylquinoline 7446-70-0, Aluminum chloride,
 reactions 7646-85-7, Zinc chloride, reactions 10136-57-9,
 5-Chloromethyl-8-Hydroxyquinoline 34825-70-2, 6-Amino-1-hexene
 81748-72-3, 5-Aminomethyl-8-Hydroxyquinoline
 (preparation of monomers and copolymers containing 8-hydroxyquinoline metal
 complexes)

L62 ANSWER 10 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2006:31888 HCPLUS Full-text
 DOCUMENT NUMBER: 144:124548
 TITLE: Use of metal complexes for immobilizing target
 molecules
 INVENTOR(S): Muir, Benjamin Ward; Barden, Michael C.; Rylatt,
 Dennis Brian; Maeji, N. Joe; Hillyard, Carmel
 Judith; Gorse, Alain-Dominique Jean-Pierre
 PATENT ASSIGNEE(S): Bio-Layer Pty. Ltd., Australia
 SOURCE: PCT Int. Appl., 118 pp.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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WO 2006002472	A1	20060112	WO 2005-AU966	20050630

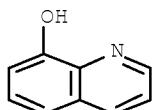
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 GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM,
 KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN,
 MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU,
 SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA,
 UG, US, UZ, VC, VN, YU, ZA, ZM, ZW

RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU,
 IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF,
 BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG,
 BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW,
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 AU 2005259833 A1 20060112 AU 2005-259833 20050630
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 JP 2008516189 T 20080515 JP 2007-519560 20050630
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 US 20090209049 A1 20090820 US 2008-571422 20080114
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 PRIORITY APPLN. INFO.: US 2004-585261P P 20040702
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 US 2005-645053P P 20050118
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 WO 2005-AU966 W 20050630
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ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

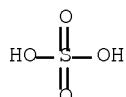
ED Entered STN: 13 Jan 2006
 AB A method of immobilizing a target mol. on a substrate, which comprises exposing the target mol. to the substrate in the presence of a metal complex, wherein the target mol. is an unmodified target mol., and wherein the metal complex is selected to provide a stable binding interaction between the target mol. and the substrate.
 IT 148-24-3D, 8-Hydroxyquinoline, complex ligand
 7440-66-6D, Zinc, complexes 10043-01-3, Alum
 (use of metal complexes for immobilizing target mols.)
 RN 148-24-3 HCPLUS
 CN 8-Quinolinol (CA INDEX NAME)



RN 7440-66-6 HCPLUS
 CN Zinc (CA INDEX NAME)

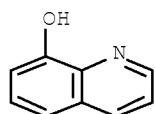
Zn

RN 10043-01-3 HCPLUS
 CN Sulfuric acid, aluminum salt (3:2) (CA INDEX NAME)

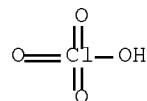


●2/3 Al

IT 148-24-3, 8-Hydroxyquinoline, reactions 10025-64-6
 (use of metal complexes for immobilizing target mols.)
 RN 148-24-3 HCPLUS
 CN 8-Quinolinol (CA INDEX NAME)



RN 10025-64-6 HCPLUS
 CN Perchloric acid, zinc salt, hydrate (2:1:6) (CA INDEX NAME)



●3 H₂O

●1/2 Zn

IC ICM C07K001-22
 ICS C07K017-06; G01N033-547
 CC 9-16 (Biochemical Methods)
 Section cross-reference(s): 1
 IT 521-31-3, Luminol
 (use of metal complexes for immobilizing target mols.)
 IT 64-19-7D, Acetic acid, salts 66-71-7D, 1,10-Phenanthroline, complex ligand 69-72-7D, Salicylic acid, complex ligand 107-15-3D, Ethylenediamine, complex ligand 110-18-9D, complex ligand 142-73-4D, Iminodiacetic acid, complex ligand 144-62-7D, Oxalic acid, complex ligand 148-24-3D, 8-Hydroxyquinoline, complex ligand 7439-89-6D, Iron, complexes 7439-96-5D, Manganese, complexes 7439-98-7D, Molybdenum, complexes 7440-02-0D, Nickel, complexes 7440-06-4D, Platinum, complexes 7440-16-6D, Rhodium, complexes 7440-18-8D, Ruthenium, complexes 7440-20-2D, Scandium, complexes 7440-32-6D, Titanium, complexes 7440-47-3D, Chromium, complexes 7440-48-4D, Cobalt, complexes 7440-50-8D, Copper,

complexes 7440-62-2D, Vanadium, complexes 7440-66-6D,
 Zinc, complexes 10043-01-3, Alum

(use of metal complexes for immobilizing target mols.)

IT 61-49-4, N- ω -Methyltryptamine 66-71-7, 1,10-Phenanthroline
 91-21-4, 1,2,3,4-Tetrahydroisoquinoline 103-67-3, Benzylmethylamine
 103-76-4, 1-(2-Hydroxyethyl)piperazine 109-01-3, 1-Methylpiperazine
 109-83-1, 2-(Methylamino)ethanol 109-85-3, 2-Methoxyethylamine
 110-68-9, N-Methylbutylamine 110-85-0, Piperazine, reactions
 110-91-8, Morpholine, reactions 111-42-2, Diethanolamine, reactions
 111-92-2, Dibutylamine 111-94-4, 3,3'-Iminodipropionitrile
 111-95-5 123-75-1, Pyrrolidine, reactions 123-90-0, Thiomorpholine
 124-40-3, Dimethylamine, reactions 139-13-9, Nitrilotriacetic acid
 142-25-6, N,N,N'-Trimethylethylenediamine 142-84-7, Dipropylamine
 148-24-3, 8-Hydroxyquinoline, reactions 504-78-9,
 Thiazolidine 589-08-2, N-Methylphenethylamine 603-35-0,
 Triphenylphosphine, reactions 622-26-4, 4-Piperidineethanol
 627-37-2, N-Methylallylamine 693-05-0,
 N-Methyl- β -alanenitrile 1066-30-4, Chromium acetate
 2439-54-5, N-Methyloctylamine 3490-06-0, N-Methylhomoveratrylamine
 4753-75-7, N-Methylfurfurylamine 5638-76-6,
 2-(2-Methylaminoethyl)pyridine 7088-89-3 7705-08-0, Iron(III)
 chloride, reactions 7720-83-4, Titanium(IV) iodide 7789-68-6,
 Titanium(IV) bromide 10025-64-6 10031-25-1, Chromium
 bromide 10031-26-2, Iron(III) bromide 10049-08-8, Ruthenium
 chloride 10101-53-8, Chromium Sulfate 10141-00-1, Chrome alum
 10294-46-9 13349-82-1 13446-57-6, Molybdenum(III) bromide
 13478-33-6, Cobalt perchlorate hexahydrate 13520-61-1, Nickel
 perchlorate hexahydrate 13537-21-8, Chromium(III) perchlorate
 13548-38-4, Chromium nitrate 13889-98-0, 1-Acetylpiperazine
 14014-88-1, Ruthenium tribromide 21359-99-9, Chromium perchlorate
 32231-06-4, 1-Piperonylpiperazine 35161-71-8, N-Methylpropargylamine
 35794-11-7, 3,5-Dimethylpiperidine 55147-94-9, Chromium perchlorate
 hexahydrate 207569-11-7 220835-52-9, Platinum iodide 320589-77-3
 698999-57-4 873196-27-1 873196-29-3

(use of metal complexes for immobilizing target mols.)

OS.CITING REF COUNT: 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS
 RECORD (1 CITINGS)

REFERENCE COUNT: 9 THERE ARE 9 CITED REFERENCES AVAILABLE FOR
 THIS RECORD. ALL CITATIONS AVAILABLE IN THE
 RE FORMAT

L62 ANSWER 11 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2005:1216411 HCAPLUS Full-text
 DOCUMENT NUMBER: 143:471965
 TITLE: Process to make metal complexes with volatile
 liquid metal compounds
 INVENTOR(S): Boone, James E.; Prindle, John C.
 PATENT ASSIGNEE(S): Albemarle Corporation, USA
 SOURCE: PCT Int. Appl., 21 pp.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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WO 2005108372	A1	20051117	WO 2004-US10505	20040405
			<--	

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA,

CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI,
 GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP,
 KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW,
 MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD,
 SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ,
 VC, VN, YU, ZA, ZM, ZW
 RW: BW, GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM,
 AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE,
 DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT,
 RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW,
 ML, MR, NE, SN, TD, TG

PRIORITY APPLN. INFO.:

WO 2004-US10505

20040405

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OTHER SOURCE(S): CASREACT 143:471965; MARPAT 143:471965

ED Entered STN: 17 Nov 2005

AB Processes for making metal complexes that are useful in organic light emitting devices (OLEDs) involve reacting a volatile liquid metal compound with a compound that has at least one hydroxyl (OH) group, such as 8-hydroxyquinoline. Exemplary volatile liquid metal compds. include trimethylaluminum, triethylaluminum, triisobutylaluminum, diisobutylaluminum hydride, dimethylaluminum chloride, and diethylaluminum chloride. One useful metal complex that can be produced is tris(8-hydroxyquinolinato)aluminum (Alq₃) via reaction of Et₃Al and 8-hydroxyquinoline in toluene at 0-40°.

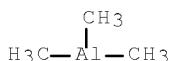
IT 75-24-1, Trimethylaluminum 97-93-8,

Triethylaluminum, reactions 148-24-3, 8-Hydroxyquinoline, reactions 148-24-30, 8-Quinolinol, substituted derivs.

(preparation of tris(hydroxyquinolinato)aluminum and related metal complexes of hydroxy-containing ligands via volatile liquid metal compound precursors)

RN 75-24-1 HCPLUS

CN Aluminum, trimethyl- (CA INDEX NAME)



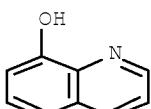
RN 97-93-8 HCPLUS

CN Aluminum, triethyl- (CA INDEX NAME)

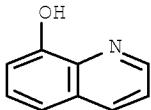


RN 148-24-3 HCPLUS

CN 8-Quinolinol (CA INDEX NAME)



RN 148-24-3 HCAPLUS
 CN 8-Quinolinol (CA INDEX NAME)



IC ICM C07D215-40
 ICS C09K011-06; C07F005-06
 CC 78-7 (Inorganic Chemicals and Reactions)
 Section cross-reference(s): 29, 73
 IT Electroluminescent devices
 Volatile substances
 (preparation of tris(hydroxyquinolinato)aluminum and related metal complexes of hydroxy-containing ligands via volatile liquid metal compound precursors)
 IT 75-24-1, Trimethylaluminum 96-10-6, Diethylaluminum chloride, reactions 97-93-8, Triethylaluminum, reactions 100-99-2, Triisobutylaluminum, reactions 148-24-3, 8-Hydroxyquinoline, reactions 148-24-3D, 8-Quinolinol, substituted derivs. 557-20-0, Diethylzinc 865-37-2, Dimethylaluminum hydride 871-27-2, Diethylaluminum hydride 1184-58-3, Dimethylaluminum chloride 1191-15-7, Diisobutylaluminum hydride 12075-68-2, Ethylaluminum sesquichloride 12542-85-7, Methylaluminum sesquichloride 18123-20-1, 4-Hydroxyacridine 71651-78-0, 3-(2-Benzothiazolyl)-4-hydroxy-2H-1-benzopyran-2-one
 (preparation of tris(hydroxyquinolinato)aluminum and related metal complexes of hydroxy-containing ligands via volatile liquid metal compound precursors)

REFERENCE COUNT: 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

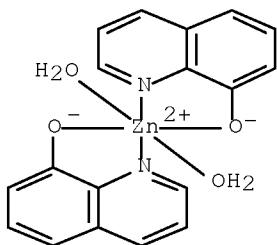
L62 ANSWER 12 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2005:1053853 HCAPLUS Full-text
 DOCUMENT NUMBER: 144:29041
 TITLE: The effects of crystal structure on optical absorption/photoluminescence of bis(8-hydroxyquinoline)zinc
 AUTHOR(S): Xu, Bing-she; Hao, Yu-ying; Wang, Hua; Zhou, He-feng; Liu, Xu-guang; Chen, Ming-wei
 CORPORATE SOURCE: College of Materials Science and Engineering, Taiyuan University of Technology, Shanxi, 030024, Peop. Rep. China
 SOURCE: Solid State Communications (2005), 136(6), 318-322
 CODEN: SSCOAA; ISSN: 0038-1098
 PUBLISHER: Elsevier Ltd.
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 ED Entered STN: 03 Oct 2005
 AB The transformation processes of 2 types of bis(8-hydroxyquinolinato)zinc: Znq2(H₂O)₂ and anhydrous (Znq2)₄ were studied by XRD, UV spectra, SEM, DSC,

and TG. The effects of crystal structure on optical properties of bis(8-hydroxyquinoiline)zinc were analyzed. Znq2(H₂O)₂ can be transformed into anhydrous (Znq2)₄ during heating under vacuum. Reversal transformation occurs by the interaction between CHCl₃ and (Znq2)₄. But (Znq2)₄ was partially transformed into Znq2 dihydrate by the interaction between EtOH and (Znq2)₄. The different mol. structure results in different crystal stacking and electronic structure, thereby affect its optical properties.

IT 15279-82-0P, DiaquaBis(8-hydroxyquinolinato)zinc
 (preparation and crystal structure effects on optical absorption/
 luminescence and thermal decomposition of)

RN 15279-82-0 HCAPLUS

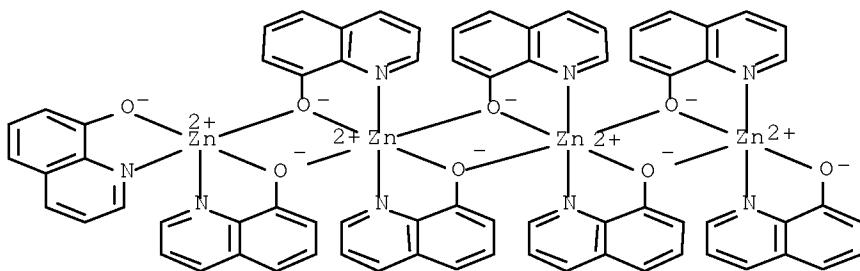
CN Zinc, diaquabis(8-quinolinolato- κ N1, κ O8)- (CA INDEX NAME)



IT 97477-20-8P
 (preparation and crystal structure effects on optical absorption/
 luminescence of)

RN 97477-20-8 HCAPLUS

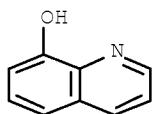
CN Zinc, hexakis[μ -(8-quinolinolato- κ N1, κ O8: κ O8)]bis(8-quinolinolato- κ N1, κ O8)tetra-, stereoisomer (9CI) (CA INDEX NAME)



IT 148-24-3, 8-Hydroxyquinoline, processes
 (reaction with zinc sulfate in presence of triethylamine)

RN 148-24-3 HCAPLUS

CN 8-Quinolinol (CA INDEX NAME)



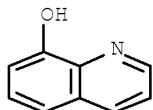
CC 73-4 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
 Section cross-reference(s): 75, 78
 ST zinc hydroxyquinoline hydrate anhyd tetramer crystal structure luminescence UV
 IT Crystal structure-property relationship (luminescence; of zinc hydroxyquinolinato complex monomer dihydrate and anhydrous tetramer)
 IT 15279-82-0P, DiaquaBis(8-hydroxyquinolinato)zinc (preparation and crystal structure effects on optical absorption/luminescence and thermal decomposition of)
 IT 97477-20-8P (preparation and crystal structure effects on optical absorption/luminescence of)
 IT 148-24-3, 8-Hydroxyquinoline, processes (reaction with zinc sulfate in presence of triethylamine)
 OS.CITING REF COUNT: 6 THERE ARE 6 CAPLUS RECORDS THAT CITE THIS RECORD (6 CITINGS)
 REFERENCE COUNT: 15 THERE ARE 15 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L62 ANSWER 13 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2005:654922 HCAPLUS [Full-text](#)
 DOCUMENT NUMBER: 143:306883
 TITLE: Poly(styrene)-Supported Alq3 and BPh2q
 AUTHOR(S): Wang, Xian-Yong; Weck, Marcus
 CORPORATE SOURCE: School of Chemistry and Biochemistry, Georgia Institute of Technology, Atlanta, GA, 30332-0400, USA
 SOURCE: Macromolecules (2005), 38(17), 7219-7224
 CODEN: MAMOBX; ISSN: 0024-9297
 PUBLISHER: American Chemical Society
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 ED Entered STN: 28 Jul 2005
 AB We describe the synthesis of 8-hydroxyquinoline-tethered poly(styrene)s as modular precursors for functionalization with metalloquinolates to form either tris(8-hydroxyquinoline)aluminum (Alq3) or 8-hydroxyquinoline biphenylboron (BPh2q) pendant polymers. All polymers, both in solution and in the solid state, show similar luminescent properties as their corresponding reference compds. Alq3 and BPh2q while retaining excellent solution-processing properties. These results clearly indicate that the poly(styrene) backbone does not interfere with the photophys. properties of the pendant Alq3 and BPh2q chromophores but endows solution processability to the materials.
 IT 97-93-8DP, Triethylaluminum, reaction products (chloromethyl)styrene-styrene copolymer derivs. and hydroxyquinoline 148-24-3DP, 8-Hydroxyquinoline, reaction products (chloromethyl)styrene-styrene copolymer derivs. and triethylaluminum (preparation and properties of poly(styrene)-supported tris(hydroxyquinoline)aluminum and hydroxyquinoline biphenylboron pendant polymers)

RN 97-93-8 HCAPLUS
 CN Aluminum, triethyl- (CA INDEX NAME)



RN 148-24-3 HCAPLUS
 CN 8-Quinolinol (CA INDEX NAME)



CC 37-3 (Plastics Manufacture and Processing)
 Section cross-reference(s): 35, 73
 IT Chromophores
 Luminescence
 (preparation and properties of poly(styrene)-supported
 tris(hydroxyquinoline)aluminum and hydroxyquinoline biphenylboron
 pendant polymers)
 IT 97-93-8DP, Triethylaluminum, reaction products
 (chloromethyl)styrene-styrene copolymer derivs. and hydroxyquinoline
 148-24-3DP, 8-Hydroxyquinoline, reaction products
 (chloromethyl)styrene-styrene copolymer derivs. and triethylaluminum
 960-71-4DP, Triphenylboron, reaction products
 (chloromethyl)styrene-styrene copolymer derivs. 1074-82-4DP,
 Potassium phthalimide, reaction products with
 (chloromethyl)styrene-styrene copolymer and formylhydroxyquinoline,
 hydrogenated, reaction products with triphenylboron or
 hydroxyquinoline and triethylaluminum 2598-30-3DP,
 5-Formyl-8-hydroxyquinoline, reaction products with
 (chloromethyl)styrene-styrene copolymer and potassium phthalimide,
 hydrogenated, reaction products with triphenylboron or
 hydroxyquinoline and triethylaluminum 29464-22-0DP,
 p-(Chloromethyl)styrene-styrene copolymer, reaction products with
 potassium phthalimide and formylhydroxyquinoline, hydrogenated,
 reaction products with triphenylboron or hydroxyquinoline and
 triethylaluminum
 (preparation and properties of poly(styrene)-supported
 tris(hydroxyquinoline)aluminum and hydroxyquinoline biphenylboron
 pendant polymers)
 OS.CITING REF COUNT: 33 THERE ARE 33 CAPLUS RECORDS THAT CITE THIS
 RECORD (33 CITINGS)
 REFERENCE COUNT: 28 THERE ARE 28 CITED REFERENCES AVAILABLE FOR
 THIS RECORD. ALL CITATIONS AVAILABLE IN THE
 RE FORMAT

L62 ANSWER 14 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2005:426178 HCAPLUS [Full-text](#)
 DOCUMENT NUMBER: 142:454113

TITLE: Organic electroluminescent devices having a stability-enhancing layer
 INVENTOR(S): Liao, Liang-Sheng; Klubek, Kevin P.
 PATENT ASSIGNEE(S): Eastman Kodak Company, USA
 SOURCE: U.S. Pat. Appl. Publ., 13 pp.
 CODEN: USXXCO
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 20050104511	A1	20050519	US 2003-713523	20031114 ---
US 7138763	B2	20061121		
WO 2005050753	A2	20050602	WO 2004-US35918	20041029 ---
WO 2005050753	A3	20050728		
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				

PRIORITY APPLN. INFO.: US 2003-713523 A 20031114

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

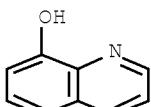
ED Entered STN: 19 May 2005

AB An organic light-emitting device with enhanced operational stability comprising an anode; a hole-transporting layer disposed over the anode; a light-emitting layer disposed over the hole-transporting layer for producing light in response to hole-electron recombination, wherein the light-emitting layer includes at least one organic host material and one organic luminescent dopant material; a stability-enhancing layer disposed in contact with the light-emitting layer, wherein the stability-enhancing layer includes at least one organic host material and one inorg. dopant material; an electron-transporting layer disposed over the stability-enhancing layer; and a cathode disposed over the electron-transporting layer.

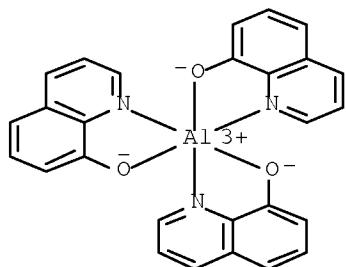
IT 148-24-3D, 8-Hydroxyquinoline, complexes 2085-33-8
, Alq3 7429-90-5, Aluminum, properties 7440-55-3
, Gallium, properties 7440-66-6, Zinc, properties
(organic electroluminescent devices having
stability-enhancing layer)

RN 148-24-3 HCPLUS

CN 8-Quinolinol (CA INDEX NAME)



RN 2085-33-8 HCAPLUS
 CN Aluminum, tris(8-quinolinolato- κ N1, κ O8)- (CA INDEX NAME)



RN 7429-90-5 HCAPLUS
 CN Aluminum (CA INDEX NAME)

Al

RN 7440-55-3 HCAPLUS
 CN Gallium (CA INDEX NAME)

Ga

RN 7440-66-6 HCAPLUS
 CN Zinc (CA INDEX NAME)

Zn

IC ICM H05B033-00
 ICS H05B033-14; H05B033-22
 INCL 313504000; 313503000; 313506000
 CC 73-12 (Optical, Electron, and Mass Spectroscopy and Other
 Related Properties)
 Section cross-reference(s): 76
 ST org electroluminescent device stability enhancing layer
 IT Electroluminescent devices
 (displays; organic electroluminescent devices having
 stability-enhancing layer)
 IT Luminescent screens
 (electroluminescent; organic
 electroluminescent devices having stability-enhancing

layer)

IT Electroluminescent devices
 (organic electroluminescent devices having
 stability-enhancing layer)

IT Alkali metals, properties
 Alkaline earth metals
 Fluoropolymers, properties
 Metals, properties
 Rare earth metals, properties
 Transition metals, properties
 (organic electroluminescent devices having
 stability-enhancing layer)

IT 148-24-3D, 8-Hydroxyquinoline, complexes 1662-01-7, Bphen
 2085-33-8, Alq3 7429-90-5, Aluminum, properties
 7429-91-6, Dysprosium, properties 7439-91-0, Lanthanum, properties
 7439-93-2, Lithium, properties 7439-94-3, Lutetium, properties
 7439-95-4, Magnesium, properties 7439-96-5, Manganese, properties
 7439-98-7, Molybdenum, properties 7440-00-8, Neodymium, properties
 7440-02-0, Nickel, properties 7440-05-3, Palladium, properties
 7440-06-4, Platinum, properties 7440-09-7, Potassium, properties
 7440-10-0, Praseodymium, properties 7440-17-7, Rubidium, properties
 7440-19-9, Samarium, properties 7440-22-4, Silver, properties
 7440-23-5, Sodium, properties 7440-24-6, Strontium, properties
 7440-27-9, Terbium, properties 7440-30-4, Thulium, properties
 7440-31-5, Tin, properties 7440-39-3, Barium, properties
 7440-45-1, Cerium, properties 7440-46-2, Cesium, properties
 7440-50-8, Copper, properties 7440-52-0, Erbium, properties
 7440-53-1, Europium, properties 7440-54-2, Gadolinium, properties
 7440-55-3, Gallium, properties 7440-57-5, Gold, properties
 7440-64-4, Ytterbium, properties 7440-65-5, Yttrium, properties
 7440-66-6, Zinc, properties 7440-70-2, Calcium, properties
 7440-74-6, Indium, properties 50926-11-9, Indium tin oxide
 80663-92-9 123847-85-8, Npb 274905-73-6
 (organic electroluminescent devices having
 stability-enhancing layer)

OS.CITING REF COUNT: 3 THERE ARE 3 CAPLUS RECORDS THAT CITE THIS
 RECORD (3 CITINGS)

REFERENCE COUNT: 28 THERE ARE 28 CITED REFERENCES AVAILABLE FOR
 THIS RECORD. ALL CITATIONS AVAILABLE IN THE
 RE FORMAT

L62 ANSWER 15 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2005:345979 HCAPLUS Full-text
 DOCUMENT NUMBER: 142:381971
 TITLE: Organic electroluminescent devices
 employing an organometallic complex-containing
 layer adjacent to a reducing metal and fabrication
 process of electroluminescent devices
 INVENTOR(S): Kido, Junji; Matsumoto, Toshio; Nakada, Takeshi;
 Kawamura, Norifumi
 PATENT ASSIGNEE(S): International Manufacturing and Engineering
 Services Co., Ltd., Japan
 SOURCE: Eur. Pat. Appl., 33 pp.
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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EP 1524707	A2	20050420	EP 2004-24611 ---<--	20041015
EP 1524707	A3	20060426		
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, PL, SK, HR				
JP 2005123094	A	20050512	JP 2003-358401 ---<--	20031017
CN 1610473	A	20050427	CN 2004-10080504 ---<--	20040930
US 20050084713	A1	20050421	US 2004-966708 ---<--	20041015
KR 2005037400	A	20050421	KR 2004-83343 ---<--	20041018
KR 858106	B1	20080910		
PRIORITY APPLN. INFO.:			JP 2003-358401 ---<--	A 20031017

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

ED Entered STN: 22 Apr 2005

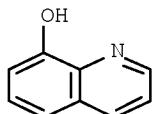
AB Organic electroluminescent devices are described which comprise a substrate, an anode layer; an organic structure including at least one light-emissive layer; a low resistance electron-transporting layer including a mixed layer of an electron-donating metal dopant and an organic compound; an organometallic complex-containing layer including an organometallic complex compound containing at least one metal ion selected from an alkaline metal ion, an alkaline earth metal ion and a rare earth metal ion; a reducing reaction generating layer; and a cathode layer, in that order. At least one of the anode layer and the cathode layer is transparent. The reducing reaction generating layer is a layer produced by depositing on the organometallic complex-containing layer a thermally reducible metal capable of reducing the metal ion in the organometallic complex compound in a vacuum to the corresponding metal, followed by causing an oxidation-reduction reaction between them.

IT 25387-93-3

(organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)

RN 25387-93-3 HCPLUS

CN 8-Quinolinol, lithium salt (1:1) (CA INDEX NAME)

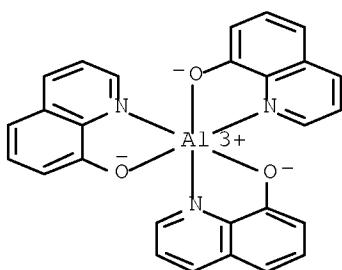


● Li

IT 2085-33-8, Aluminum tris(8-hydroxyquinolinato)
(organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)

RN 2085-33-8 HCPLUS

CN Aluminum, tris(8-quinolinolato-κN₁,κO₈) - (CA INDEX NAME)



IT 7429-90-5, Aluminum, uses
 (thermally reducible metal; organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)

RN 7429-90-5 HCPLUS

CN Aluminum (CA INDEX NAME)

Al

IC ICM H01L051-20

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
 Section cross-reference(s): 76

ST OLED organometallic complex reducing metal fabrication; org electroluminescent device organometallic complex reducing metal manuf

IT Organometallic compounds
 (alkaline earth compds.; organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)

IT Organometallic compounds
 (alkali metal compds.; organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)

IT Sputtering
 (cathode deposition by; organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)

IT Vapor deposition process
 (electron-beam, thermally reducible metal deposition by; organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)

IT Vapor deposition process
 (laser ablation, thermally reducible metal deposition by; organic electroluminescent devices employing organometallic

- complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)
- IT Electroluminescent devices
 Semiconductor device fabrication
 (organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)
- IT Coordination compounds
 Organometallic compounds
 (organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)
- IT Alkali metals, uses
 Alkaline earth metals
 Rare earth metals, uses
 (organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal resulting in formation of)
- IT Alkali metal compounds
 Alkaline earth compounds
 Rare earth compounds
 (organometallic compds.; organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)
- IT Organometallic compounds
 (rare earth compds.; organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)
- IT Vapor deposition process
 (resistive heating; organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)
- IT Reduction
 (thermal; organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)
- IT 50926-11-9, Indium tin oxide
 (electrode; organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)
- IT 123847-85-8, α -NPD
 (hole-transporting layer; organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)
- IT 25387-93-3
 (organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)
- IT 2085-33-8, Aluminum tris(8-hydroxyquinolinato) 4733-39-5,
 Bathocuproine 7440-46-2, Cesium, uses
 (organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication process of electroluminescent devices)
- IT 7439-93-2P, Lithium, uses
 (organic electroluminescent devices employing organometallic complex-containing layer adjacent to reducing metal and fabrication

process of electroluminescent devices)
IT 7429-90-S, Aluminum, uses
(thermally reducible metal; organic electroluminescent
devices employing organometallic complex-containing layer adjacent to
reducing metal and fabrication process of
electroluminescent devices)

IT 7440-21-3, Silicon, uses 7440-32-6, Titanium, uses 7440-33-7,
Tungsten, uses 7440-67-7, Zirconium, uses
(thermally reducible metal; organic electroluminescent
devices employing organometallic complex-containing layer adjacent to
reducing metal and fabrication process of
electroluminescent devices)

OS.CITING REF COUNT: 2 THERE ARE 2 CAPLUS RECORDS THAT CITE THIS
RECORD (8 CITINGS)

REFERENCE COUNT: 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR
THIS RECORD. ALL CITATIONS AVAILABLE IN THE
RE FORMAT

L62 ANSWER 16 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN

ACCESSION NUMBER: 2005:119243 HCPLUS Full-text

DOCUMENT NUMBER: 142:373950

TITLE: Synthesis, Characterization, and Photophysical
Properties of Iridium Complexes with an
8-Phenylquinoline Framework. The First
Six-Membered Chelated Iridium Complexes for
Electroluminescence

AUTHOR(S): Li, Hao-Chun; Chou, Pi-Tai; Hu, Ya-Hui; Cheng,
Yi-Ming; Liu, Rai-Shung

CORPORATE SOURCE: Department of Chemistry, National Tsing-Hua
University, Hsinchu, Taiwan

SOURCE: Organometallics (2005), 24(6), 1329-1335
CODEN: ORGND7; ISSN: 0276-7333

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal

LANGUAGE: English

OTHER SOURCE(S): CASREACT 142:373950

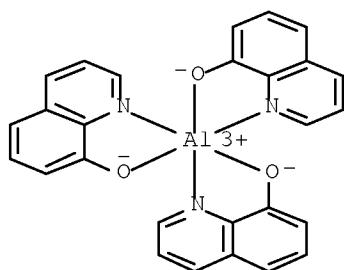
ED Entered STN: 11 Feb 2005

AB Six-membered chelated Ir complexes bearing an 8-phenylquinoline framework were prepared and characterized with an x-ray diffraction study. The photophys. properties of these complexes were examined with appropriate spectroscopic methods. The results, in combination with ab initio approaches, led the authors to clearly assign various electronic transition states. One salient feature for these red complexes is the appearance of a dual strong absorption band around 425-500 nm, which incorporates a state mixing between 1MLCT and $\pi\pi^*$ manifolds. These complexes show deep red phosphorescent emissions (650-680 nm), with short lifetimes of 1.7-3.0 μ s and moderate quantum yields of 0.05-0.11 in deaerated MeCN. The electroluminescence performance of these species shows a promising perspective in the OLED display. One representative, Ir[8-(3,5-difluorophenyl)quinoline]2(acac) (4c), shows an η_{ext} value of 2.04% at J = 20 mA/cm² and the maximum brightness is 3427 cd/m² ($x = 0.68$, $y = 0.32$) with a full width at half-maximum of only 60 nm at 16 V, demonstrating the 1st six-membered chelated Ir complexes to suit applications in OLED devices.

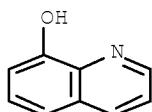
IT 2085-33-8, Tris(8-hydroxyquinoline)aluminum
(preparation and photophys. properties of cyclometalated iridium
complexes with phenylquinoline framework and six-membered chelated
ring with acetylacetone)

RN 2085-33-8 HCPLUS

CN Aluminum, tris(8-quinolinolato- κ N1, κ O8)- (CA INDEX NAME)



IT 148-24-3, 8-Hydroxyquinoline, reactions
 (preparation and photophys. properties of cyclometalated iridium complexes with phenylquinoline framework and six-membered chelated ring with acetylacetone)
 RN 148-24-3 HCAPLUS
 CN 8-Quinolinol (CA INDEX NAME)



CC 29-13 (Organometallic and Organometalloidal Compounds)
 Section cross-reference(s): 22, 73, 74, 75
 ST iridium phenylquinoline cyclometalated prepn photophys electroluminescence; crystal structure iridium cyclometalated phenylquinoline fluoro prepn; mol structure iridium cyclometalated phenylquinoline fluoro
 IT Electroluminescent devices
 (organic LEDs; preparation and photophys. properties of cyclometalated iridium complexes with phenylquinoline framework and six-membered chelated ring with acetylacetone)
 IT 2085-33-8, Tris(8-hydroxyquinoline)aluminum 4733-39-5,
 Bathocuproine 50926-11-9, ITO 58328-31-7 123847-85-8
 (preparation and photophys. properties of cyclometalated iridium complexes with phenylquinoline framework and six-membered chelated ring with acetylacetone)
 IT 123-54-6, Acetylacetone, reactions 148-24-3,
 8-Hydroxyquinoline, reactions
 (preparation and photophys. properties of cyclometalated iridium complexes with phenylquinoline framework and six-membered chelated ring with acetylacetone)
 OS.CITING REF COUNT: 35 THERE ARE 35 CAPLUS RECORDS THAT CITE THIS RECORD (35 CITINGS)
 REFERENCE COUNT: 52 THERE ARE 52 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L62 ANSWER 17 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2005:54153 HCAPLUS Full-text
 DOCUMENT NUMBER: 142:280485

TITLE: Synthesis and photophysical characterization of
 the free-radical copolymerization of
 metaloquinolate-pendant monomers with methyl
 methacrylate

AUTHOR(S): Du, Naiying; Tian, Renyu; Peng, Junbiao; Lu,
 Mangeng

CORPORATE SOURCE: Key Laboratory of Polymer Materials for
 Electronics, Guangdong Guangzhou Institute of
 Chemistry, Chinese Academy of Sciences, Guangzhou,
 510650, Peop. Rep. China

SOURCE: Journal of Polymer Science, Part A: Polymer
 Chemistry (2004), Volume Date 2005,
 43(2), 397-406

CODEN: JPACEC; ISSN: 0887-624X

PUBLISHER: John Wiley & Sons, Inc.

DOCUMENT TYPE: Journal

LANGUAGE: English

ED Entered STN: 20 Jan 2005

AB A series of novel metaloquinolate [aluminum quinolate (Alq3), zinc quinolate (Znq2), and lithium quinolate (Liq)]-containing copolymers were prepared. This is the first report of the synthesis of metaloquinolate-containing polymers by free-radical copolymer. The structures of the metaloquinolate monomers and copolymers were characterized by ¹H NMR and Fourier-transform IR techniques. The differential scanning calorimetry and thermogravimetric anal. results showed that the copolymers were more thermally stable than the Me methacrylate homopolymer. The copolymers (<25 weight % Alq3, <20 weight % Znq2, or <15 weight % Liq) could be dissolved in common solvents without crosslinking. The UV-visible absorption and photoluminescence (PL) emission properties of the copolymers were consistent with the literature data of metaloquinolate complexes. The PL efficiencies of the metaloquinolate-containing copolymers with 25 weight % Alq3, 20 weight % Znq2, or 15 weight % Liq were 19.89, 13.24, and 11.82%, resp. The exptl. results indicated that these kinds of materials could be used for practical applications in organic light-emitting diodes.

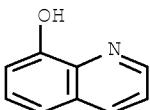
IT 97-93-8, Triethylaluminum, reactions 148-24-3,
 8-Quinolinol, reactions
 (in synthesis of hydroxyquinoline metal chelates for free-radical
 polymerization with Me methacrylate)

RN 97-93-8 HCAPLUS

CN Aluminum, triethyl- (CA INDEX NAME)



RN 148-24-3 HCAPLUS
 CN 8-Quinolinol (CA INDEX NAME)



CC 35-4 (Chemistry of Synthetic High Polymers)
 ST metaloquinolate contg monomer radical polymn methyl methacrylate; zinc
 quinolate contg polymethacrylate prepn property; aluminum quinolate
 contg polymethacrylate prepn property; lithium quinolate contg
 polymethacrylate prepn property; light emitting
 diode metaloquinolate contg polymethacrylate
 IT Electroluminescent devices
 (synthesis and photophys. characterization of copolymers of
 metaloquinolate-pendant monomers and Me methacrylate for use in)
 IT Glass transition temperature
 Luminescence
 Molecular weight
 Polydispersity
 Solubility
 Thermal stability
 UV and visible spectra
 (synthesis and photophys. characterization of free-radical
 copolymn. of metaloquinolate-pendant monomers with Me methacrylate)
 IT 50-00-0, Formaldehyde, reactions 97-93-8,
 Triethylaluminum, reactions 123-31-9, Hydroquinol, reactions
 148-24-3, 8-Quinolinol, reactions 557-20-0, Diethylzinc
 868-77-9, 2-Hydroxyethyl methacrylate 1310-65-2, Lithium hydroxide
 (in synthesis of hydroxyquinoline metal chelates for free-radical
 polymerization with Me methacrylate)
 OS.CITING REF COUNT: 16 THERE ARE 16 CAPLUS RECORDS THAT CITE THIS
 RECORD (16 CITINGS)
 REFERENCE COUNT: 28 THERE ARE 28 CITED REFERENCES AVAILABLE FOR
 THIS RECORD. ALL CITATIONS AVAILABLE IN THE
 RE FORMAT

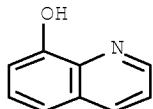
L62 ANSWER 18 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2004:1080334 HCPLUS Full-text
 DOCUMENT NUMBER: 142:47531
 TITLE: Method of reducing photoelectric device leakage
 current in conjugated polymer, and conjugated
 polymer composition
 INVENTOR(S): Chen, Kuo-Yu; Tuan, Chi-Shen; Teng, Wan-Jung;
 Chang, Shinn-Jen
 PATENT ASSIGNEE(S): Industrial Technology Research Institute, Taiwan
 SOURCE: U.S. Pat. Appl. Publ., 17 pp.
 CODEN: USXXCO
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
-----	----	-----	-----	-----
US 20040250849	A1	20041216	US 2003-633708 <--	20030805
US 7026275	B2	20060411		
TW 255832	B	20060601	TW 2003-92115804 <--	20030611
PRIORITY APPLN. INFO.:			TW 2003-92115804 <--	A 20030611

ED Entered STN: 17 Dec 2004
 AB A method of reducing the photoelec. device leakage current caused by residual
 metal ions in conjugated polymer. A chelating agent is added to a conjugated
 polymer material, thereby the conductivity and mobility of metal ions under an
 elec. field are reduced due to the chelation of metal ions by the chelating

agent; therefore, the leakage current is reduced and the stability of devices is improved. Also, the activity of metal ions is reduced after the metal ions are chelated by the chelating agent, improving the stability of the material and the devices. A conjugated polymer composition is also provided.

- IT 148-24-3, 8-Hydroxyquinoline, processes
 (chelating agent; fabricating method of reducing photoelec. device leakage current in conjugated polymer by addition of chelating agents and conjugated polymer composition)
- RN 148-24-3 HCPLUS
- CN 8-Quinolinol (CA INDEX NAME)



- IT 7429-90-5, Aluminum, uses
 (contact; fabricating method of reducing photoelec. device leakage current in conjugated polymer by addition of chelating agents and conjugated polymer composition)
- RN 7429-90-5 HCPLUS
- CN Aluminum (CA INDEX NAME)

Al

- IC ICM H01L021-00
 INCL 136263000; 438149000; 438151000; 438057000; 438059000; 438463000
 CC 76-5 (Electric Phenomena)
 Section cross-reference(s): 38, 48, 52, 74
- IT Capacitors
 Chelating agents
 Electroluminescent devices
 Gettering
 Inductors
 Lasers
 Memory devices
 Photoelectric devices
 Resistors
 Solar cells
 Solvents
 Transistors
 (fabricating method of reducing photoelec. device leakage current in conjugated polymer by addition of chelating agents and conjugated polymer composition)
- IT 94-93-9, N,N'-Bis(salicylidene)ethylenediamine 148-24-3,
 8-Hydroxyquinoline, processes
 (chelating agent; fabricating method of reducing photoelec. device leakage current in conjugated polymer by addition of chelating agents and conjugated polymer composition)
- IT 7429-90-5, Aluminum, uses 7440-70-2, Calcium, uses
 (contact; fabricating method of reducing photoelec. device leakage

current in conjugated polymer by addition of chelating agents and conjugated polymer composition)

OS.CITING REF COUNT: 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1 CITINGS)
 REFERENCE COUNT: 1 THERE ARE 1 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L62 ANSWER 19 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2004:1059398 HCAPLUS Full-text
 DOCUMENT NUMBER: 142:24413
 TITLE: Luminescent organic-polymer/metal complex for wet processable films and process for producing thereof
 INVENTOR(S): Kudo, Kazuaki; Kobayashi, Yasushi; Takayama, Toshio; Sano, Hiroshi
 PATENT ASSIGNEE(S): Nippon Light Metal Company, Ltd., Japan
 SOURCE: PCT Int. Appl., 47 pp.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2004106389	A1	20041209	WO 2004-JP7904	20040601
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W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
JP 2005015752	A	20050120	JP 2003-311117	20030903
			<--	
JP 4295047	B2	20090715		
EP 1637545	A1	20060322	EP 2004-735677	20040601
			<--	
EP 1637545	B1	20091111		
R: DE, GB				
CN 1798776	A	20060705	CN 2004-80014974	20040601
			<--	
CN 100503655	C	20090624		
US 20060128940	A1	20060615	US 2005-559129	20051201
			<--	
PRIORITY APPLN. INFO.:			JP 2003-156790	A 20030602
			<--	
			JP 2003-311117	A 20030903
			<--	
			WO 2004-JP7904	W 20040601
			<--	

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT
 ED Entered STN: 10 Dec 2004

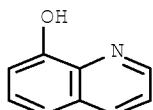
AB The present invention relates to a luminescent org .-polymer/metal complex polyQ·M·Kn-1 obtained by reacting a polymer ligand (polyQ) comprising a polymer and a side chain 8-hydroxyquinoline derivative through a spacer group, a 8-hydroxyquinoline derivative ligand (K), and a di- to tetravalent metal ion (Mn⁺), wherein the 8-hydroxyquinoline derivative ligand has a bulky substituent. Also provided is a luminescent org .-polymer/metal complex composition comprising the luminescent organic-polymer/metal complex and a low-mol. ligand. Thus, 26.2 g 8-hydroxyquinoline and hydro chloride were reacted in the presence of formaldehyde, 2-hydroxyethyl methacrylate was added therein and reacted to give 2-ethoxy methacrylate-substituted 8-hydroxyquinoline, 7.00 g of which was polymerized in the presence of 321 mg AIBN to give a polymer ligand, 800 mg of which was mixed with 6.64 g 7-(4-ethyl-1-methyloctyl)-8-hydroxyquinoline and reacted with 8.4 mmol trimethylaluminum to give a aluminum complex mixture showing luminescence at 400 nm (excitation at 533 nm) for both in solution and in film.

IT 148-24-3, 8-Hydroxyquinoline, reactions

(reactant in polymer ligand preparation; preparation of luminescent organic-polymer/metal complexes for wet processable films)

RN 148-24-3 HCPLUS

CN 8-Quinolinol (CA INDEX NAME)



IT 75-24-1, Trimethylaluminum

(reactant in polymer ligand-metal complex preparation; preparation of luminescent organic-polymer/metal complexes for wet processable films)

RN 75-24-1 HCPLUS

CN Aluminum, trimethyl- (CA INDEX NAME)



IC ICM C08F008-42

ICS C08F020-36; C09K011-06; H05B033-14

CC 38-3 (Plastics Fabrication and Uses)

Section cross-reference(s): 73

ST luminescent org polymer metal complex wet processable film process; hydroxyquinoline hydro chloride formaldehyde reactant; aluminum polymeric ligand complex prepn

IT Coordination compounds

(polymer ligand-; preparation of luminescent org .-polymer/metal complexes for wet processable films)

IT Ligands

(polymeric; preparation of luminescent org .-polymer/metal complexes for wet processable films)

IT Luminescent substances

(preparation of luminescent organic-polymer/metal complexes for wet processable films)

IT 4053-45-6P 745782-29-0P 745782-35-8P 745782-37-0P
 (intermediate in polymer ligand preparation; preparation of luminescent organic-polymer/metal complexes for wet processable films)

IT 84953-68-4
 (ligand; preparation of luminescent org.-polymer/metal complexes for wet processable films)

IT 95831-48-4P 745782-31-4P
 (monomer in polymer ligand preparation; preparation of luminescent organic-polymer/metal complexes for wet processable films)

IT 745782-32-5P 802985-76-8P
 (polymer ligand; preparation of luminescent org.-polymer/metal complexes for wet processable films)

IT 7429-90-5DP, Aluminum, polymer complexes 73545-11-6DP, aluminum complexes 745782-32-5DP, aluminum complexes 802985-76-8DP, aluminum complexes
 (preparation of luminescent organic-polymer/metal complexes for wet processable films)

IT 18162-48-6, tert-Butyldimethylsilyl chloride
 (protecting group in polymer ligand preparation; preparation of luminescent organic-polymer/metal complexes for wet processable films)

IT 50-00-0, Formaldehyde, reactions 148-24-3,
 8-Hydroxyquinoline, reactions 868-77-9, 2-Hydroxyethyl methacrylate
 7647-01-0, Hydrochloric acid, reactions 73545-11-6
 (reactant in polymer ligand preparation; preparation of luminescent organic-polymer/metal complexes for wet processable films)

IT 75-24-1, Trimethylaluminum 555-31-7, Triisopropoxyaluminum
 (reactant in polymer ligand-metal complex preparation; preparation of luminescent organic-polymer/metal complexes for wet processable films)

REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L62 ANSWER 20 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2004:1037594 HCPLUS Full-text
 DOCUMENT NUMBER: 142:206760
 TITLE: Self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices
 AUTHOR(S): Kwok, Chi-Chung; Yu, Sze-Chit; Sham, Iona H. T.; Che, Chi-Ming
 CORPORATE SOURCE: Department of Chemistry and the HKU-CAS Joint Laboratory on New Materials, The University of Hong Kong, Hong Kong, Peop. Rep. China
 SOURCE: Chemical Communications (Cambridge, United Kingdom) (2004), (23), 2758-2759
 CODEN: CHCOFS; ISSN: 1359-7345
 PUBLISHER: Royal Society of Chemistry
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 OTHER SOURCE(S): CASREACT 142:206760
 ED Entered STN: 03 Dec 2004
 AB Thermally stable zinc(II) Schiff base polymers (decomposition temperature up to 461°; Mn = 13580 to 20440) formed by self-assembly reactions of zinc(II)

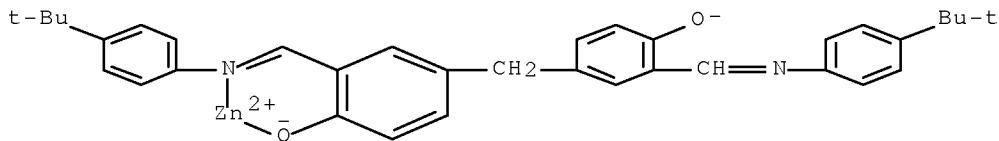
salts and salicylaldimine monomers exhibit blue to yellow PL with quantum yields up to 0.34 in DMF; PLEDs employing these polymers as emitters give green or orange EL with turn-on voltage at 5 and 6 V and maximum efficiency of 2.0 and 2.6 cd A⁻¹ resp.

IT 837411-92-4P 837411-96-8P

(polymeric; self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices)

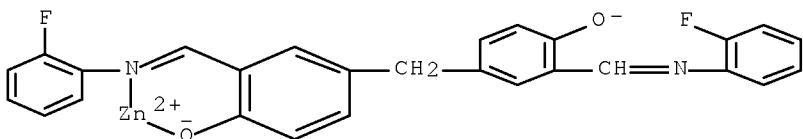
RN 837411-92-4 HCAPLUS

CN Zinc, [2-[[[4-(1,1-dimethylethyl)phenyl]imino- κ N]methyl]-4-[[3-[[[4-(1,1-dimethylethyl)phenyl]imino]methyl]-4-hydroxyphenyl]methyl]phenolato(2-)- κ O]- (CA INDEX NAME)



RN 837411-96-8 HCAPLUS

CN Zinc, [2-[[2-fluorophenyl]imino- κ N]methyl]-4-[[3-[[2-fluorophenyl]imino]methyl]-4-hydroxyphenyl]methyl]phenolato(2-)- κ O]- (CA INDEX NAME)

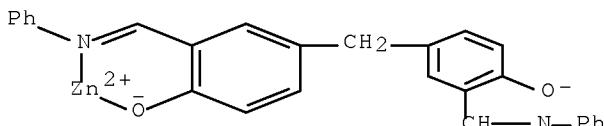


IT 75785-68-1P 837411-91-3P

(polymeric; self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices)

RN 75785-68-1 HCAPLUS

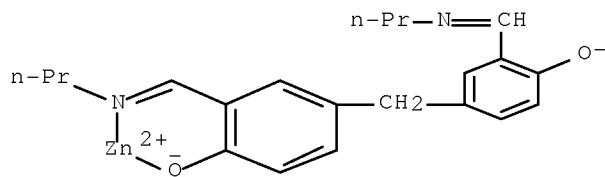
CN Zinc, [4-[[4-hydroxy-3-[(phenylimino)methyl]phenyl]methyl]-2-[(phenylimino- κ N)methyl]phenolato(2-)- κ O]- (CA INDEX NAME)



RN 837411-91-3 HCAPLUS

CN Zinc, [4-[[4-hydroxy-3-[(propylimino)methyl]phenyl]methyl]-2-[(propylimino- κ N)methyl]phenolato(2-)- κ O]- (CA INDEX

NAME)

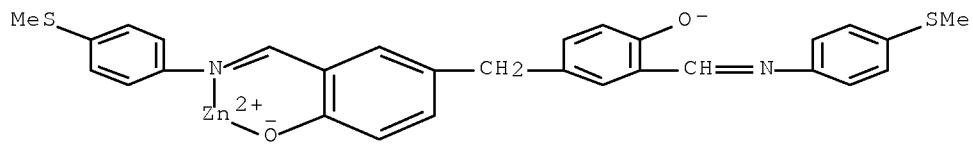


IT 837411-93-5P 837411-94-6P 837411-95-7P
 837411-97-9P

(polymeric; self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices)

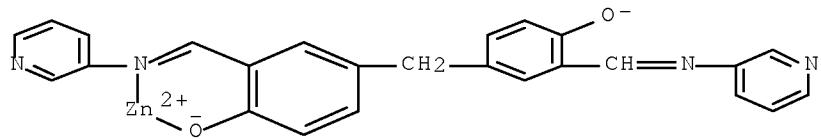
RN 837411-93-5 HCAPLUS

CN Zinc, [4-[4-hydroxy-3-[[[4-(methylthio)phenyl]imino]methyl]phenyl]methyl]-2-[[[4-(methylthio)phenyl]imino- κ N]methyl]phenolato(2-)- κ O]- (CA INDEX NAME)



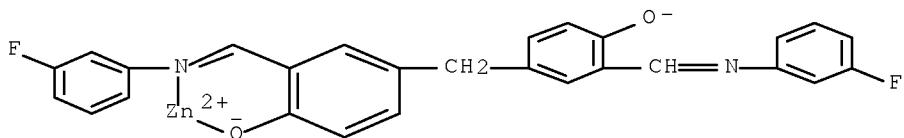
RN 837411-94-6 HCAPLUS

CN Zinc, [4-[4-hydroxy-3-[(3-pyridinylimino)methyl]phenyl]methyl]-2-[(3-pyridinylimino- κ N)methyl]phenolato(2-)- κ O]- (CA INDEX NAME)

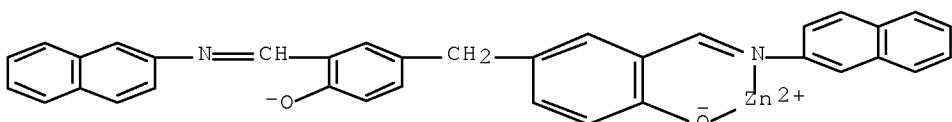


RN 837411-95-7 HCAPLUS

CN Zinc, [2-[(3-fluorophenyl)imino- κ N]methyl]-4-[[3-[(3-fluorophenyl)imino]methyl]-4-hydroxyphenyl]methyl]phenolato(2-)- κ O]- (CA INDEX NAME)

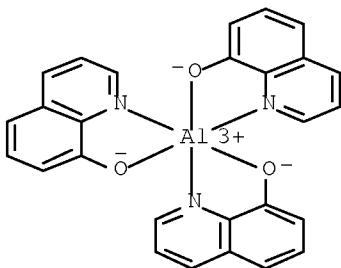


RN 837411-97-9 HCAPLUS
 CN Zinc, [4-[4-hydroxy-3-[(2-naphthalenylimino)methyl]phenyl]methyl]-2-[(2-naphthalenylimino- κ N)methyl]phenolato(2-)- κ O]- (CA INDEX NAME)



IT 2085-33-8, Alq3
 (self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices containing)

RN 2085-33-8 HCAPLUS
 CN Aluminum, tris(8-quinolinolato- κ N1, κ O8)- (CA INDEX NAME)



CC 73-5 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
 Section cross-reference(s): 36, 38, 76
 ST self assembled zinc Schiff base polymer electroluminescent device photoluminescence
 IT Polymers
 (Schiff base-containing, zinc; self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices)
 IT Luminescence, electroluminescence
 Surface structure
 Thermal stability
 (of self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices)

IT Electroluminescent devices
 (polymer; self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices)

IT Electric current-potential relationship
 Self-assembly
 (self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices)

IT Luminescence
 (visible; of self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices)

IT 9003-53-6D, sulfonated
 (dopant for PEDOT; self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices containing)

IT 126213-51-2
 (doped with PSS; self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices containing)

IT 837411-92-4P 837411-96-8P
 (polymeric; self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices)

IT 75785-68-1P 837411-91-3P
 (polymeric; self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices)

IT 837411-93-5P 837411-94-6P 837411-95-7P
 837411-97-9P
 (polymeric; self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices)

IT 62-53-3, Benzenamine, properties 91-59-8, 2-Naphthalenamine
 348-54-9 372-19-0 769-92-6 1783-81-9 3046-82-0
 (self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices)

IT 462-08-8P, 3-Pyridinamine 17867-16-2P 20880-14-2P 835897-13-7P
 835897-14-8P 835897-15-9P 835897-16-0P 835897-17-1P
 835897-18-2P
 (self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices)

IT 2085-33-8, Alq3 4733-39-5, BCP
 (self-assembled zinc(II) Schiff base polymers for applications in polymer light-emitting devices containing)

OS.CITING REF COUNT: 13 THERE ARE 13 CAPLUS RECORDS THAT CITE THIS RECORD (13 CITINGS)

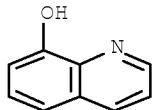
REFERENCE COUNT: 16 THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L62 ANSWER 21 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2004:1032472 HCAPLUS Full-text
 DOCUMENT NUMBER: 142:187779
 TITLE: Programmable polymer thin film and non-volatile memory device
 AUTHOR(S): Ouyang, Jianyong; Chu, Chih-Wei; Szmarda, Charles R.; Ma, Liping; Yang, Yang
 CORPORATE SOURCE: Department of Materials Science and Engineering, University of California, Los Angeles, CA, 90095, USA
 SOURCE: Nature Materials (2004), 3(12), 918-922
 CODEN: NMAACR; ISSN: 1476-1122
 PUBLISHER: Nature Publishing Group
 DOCUMENT TYPE: Journal

LANGUAGE: English
 ED Entered STN: 02 Dec 2004
 AB Building on the success of organic electronic devices, such as light-emitting diodes and field-effect transistors, procedures for fabricating nonvolatile organic memory devices are now being explored. We demonstrate an organic memory device fabricated by solution processing. Programmable elec. bistability was observed in a device made from a polystyrene film containing gold nanoparticles and 8-hydroxyquinoline sandwiched between two metal electrodes. The as-prepared device, which is in a low-conductivity state, displays an abrupt transition to a high-conductivity state under an external bias of 2.8 V. These two states differ in conductivity by about four orders of magnitude. Applying a neg. bias of 1.8 V causes the device to return to the low-conductivity state. The electronic transition is attributed to the elec.-field-induced charge transfer between the gold nanoparticles and 8-hydroxyquinoline. The transition from the low- to the high-conductivity state takes place in nanoseconds, and is nonvolatile, indicating that the device may be used for low-cost, high-d. memory storage.
 IT 7429-90-5, Aluminum, uses
 (electrodes; programmable polymer thin film and nonvolatile memory device)
 RN 7429-90-5 HCPLUS
 CN Aluminum (CA INDEX NAME)

A1

IT 148-24-3, 8-Hydroxyquinoline, uses
 (programmable polymer thin film and nonvolatile memory device)
 RN 148-24-3 HCPLUS
 CN 8-Quinolinol (CA INDEX NAME)



CC 76-14 (Electric Phenomena)
 Section cross-reference(s): 38
 IT 7429-90-5, Aluminum, uses
 (electrodes; programmable polymer thin film and nonvolatile memory device)
 IT 148-24-3, 8-Hydroxyquinoline, uses 9003-53-6, Polystyrene
 (programmable polymer thin film and nonvolatile memory device)
 OS.CITING REF COUNT: 247 THERE ARE 247 CAPLUS RECORDS THAT CITE THIS RECORD (247 CITINGS)
 REFERENCE COUNT: 25 THERE ARE 25 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L62 ANSWER 22 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2004:823958 HCPLUS Full-text
 DOCUMENT NUMBER: 141:340538

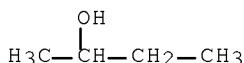
TITLE: Organic/inorganic hybrid material, composition for synthesizing the same, and process for producing the hybrid material
 INVENTOR(S): Seo, Satoshi; Nakashima, Harue; Nomura, Ryoji
 PATENT ASSIGNEE(S): Semiconductor Energy Laboratory Co. Ltd., Japan
 SOURCE: PCT Int. Appl., 99 pp.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2004085543	A1	20041007	WO 2004-JP3610	20040317 <--
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
RW: BW, GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
EP 1607446	A1	20051221	EP 2004-721373	20040317 <--
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, PL, SK				
CN 1795241	A	20060628	CN 2004-80014783	20040317 <--
CN 100422269	C	20081001		
CN 101368006	A	20090218	CN 2008-10210659	20040317 <--
US 20040265253	A1	20041230	US 2004-809130	20040325 <--
US 7517470	B2	20090414		
US 20090206746	A1	20090820	US 2009-420548	20090408 <--
PRIORITY APPLN. INFO.:			JP 2003-85688	A 20030326 <--
			CN 2004-80014783	A3 20040317 <--
			WO 2004-JP3610	W 20040317 <--
			US 2004-809130	A3 20040325 <--

- ED Entered STN: 08 Oct 2004
 AB An organic/inorg. hybrid material contains an organic compound (functional chelating agent) which is capable of manifesting coloring, luminescence and semiconductor property functions through formation of a chelate with metal atom and is in pendant-form chelate coordination with the metal atom of metal oxide matrix. This organic/inorg. hybrid material can be synthesized by preparing a sol (coating composition) comprising a metal alkoxide and/or metal salt and a functional chelating agent and thereafter carrying out synthetic operation according to sol gel process. Thus, there can be synthesized an

organic/inorg. hybrid material which has an organic group directly linked to a metal oxide matrix and consequently can manifest functions different from those of simple metal oxide. That is, a functional organic/inorg. hybrid material capable of manifesting coloring, luminescence or semiconductor properties can be realized by the organic group directly linked to a metal oxide matrix.

IT 2269-22-9 7446-70-0, Aluminum chloride,
processes
(organic/inorg. hybrid material having coloring, luminescence
or semiconductor properties)
RN 2269-22-9 HCAPLUS
CN 2-Butanol, aluminum salt (3:1) (CA INDEX NAME)

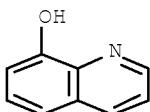


● 1 / 3 Al

RN 7446-70-0 HCAPLUS
CN Aluminum chloride (AlCl₃) (CA INDEX NAME)



IT 148-24-3DP, 8-Quinolinol, chelates with metal oxide matrix
1344-28-1P, Alumina, preparation
(organic/inorg. hybrid material having coloring, luminescence
or semiconductor properties)
RN 148-24-3 HCAPLUS
CN 8-Quinolinol (CA INDEX NAME)



RN 1344-28-1 HCPLUS
CN Aluminum oxide (Al₂O₃) (CA INDEX NAME)

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

IC ICM C09B057-10
 ICS C09K011-06; C09D185-00; C09D001-00; C09D007-12; H05B033-14;
 C03C017-32

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 42

ST org inorg hybrid coloring luminescence semiconductor
property

IT Metal alkoxides
(coating; organic/inorg. hybrid material having coloring,
luminescence or semiconductor properties)

IT Electroluminescent devices
(displays; organic/inorg. hybrid material having coloring,
luminescence or semiconductor properties)

IT Luminescent screens
(electroluminescent; organic/inorg. hybrid
material having coloring, luminescence or semiconductor
properties)

IT Coating materials
Coloring materials
Luminescent substances
Semiconductor materials
(organic/inorg. hybrid material having coloring,
luminescence or semiconductor properties)

IT Oxides (inorganic), uses
(organic/inorg. hybrid material having coloring, luminescence
or semiconductor properties)

IT 78-10-4D, Tetraethoxysilane, hydrolyzed 2269-22-9
7446-70-0, Aluminum chloride, processes
(organic/inorg. hybrid material having coloring, luminescence
or semiconductor properties)

IT 148-24-3DP, 8-Quinolinol, chelates with metal oxide matrix
1344-28-1P, Alumina, preparation 7631-86-9P, Silica,
preparation
(organic/inorg. hybrid material having coloring, luminescence
or semiconductor properties)

IT 141-97-9, Ethyl acetoacetate
(organic/inorg. hybrid material having coloring, luminescence
or semiconductor properties)

OS.CITING REF COUNT: 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS
RECORD (1 CITINGS)

REFERENCE COUNT: 11 THERE ARE 11 CITED REFERENCES AVAILABLE FOR
THIS RECORD. ALL CITATIONS AVAILABLE IN THE
RE FORMAT

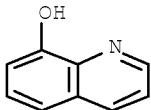
L62 ANSWER 23 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
ACCESSION NUMBER: 2004:702156 HCAPLUS Full-text
DOCUMENT NUMBER: 141:226346
TITLE: Metal 8-hydroxyquinoline-functionalized polymers
and their preparation for optical materials
INVENTOR(S): Weck, Marcus; Meyers, Amy
PATENT ASSIGNEE(S): Georgia Tech Research Corporation, USA
SOURCE: PCT Int. Appl., 88 pp.
CODEN: PIXXD2
DOCUMENT TYPE: Patent
LANGUAGE: English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2004073030	A2	20040826	WO 2004-US3587	20040206 <--
WO 2004073030	A3	20040923		
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI,				

GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP,
 KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW,
 MX, MZ, NA, NI
 RW: BW, GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT,
 BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,
 IT, LU, MC, NL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI,
 CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
 US 20050131175 A1 20050616 US 2004-773980 20040206
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 US 7105617 B2 20060912 US 2003-445701P P 20030206
 PRIORITY APPLN. INFO.: <--
 US 2003-500000P P 20030904
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 OTHER SOURCE(S): MARPAT 141:226346
 ED Entered STN: 27 Aug 2004
 AB This invention relates to the synthesis of Mqn-containing monomeric compds., comprising a polymerizable moiety, an Mqn-moiety, and an optional chemical spacer therebetween, wherein q, in each instance, comprises an 8-hydroxyquinoline residue, M is a metal such as Mg, Zn, Al, Ga, or In, and n is 2 or 3 as the valence of the metal requires. For example, the polymerization of Znq2- or Alq3-containing monomers, in the presence or absence of a co-monomer, provided a Znq2- or Alq3-containing polymer, which retained the optical properties of Znq2 or Alq3 in solution, resp. The Mqn-containing polymers may be used in, among other things, the fabrication of light-emitting diodes (LEDs).
 IT 97-93-8, Triethylaluminum, reactions 148-24-3,
 8-Hydroxyquinoline, reactions (metal 8-hydroxyquinoline-functionalized polymers and their preparation for optical materials)
 RN 97-93-8 HCPLUS
 CN Aluminum, triethyl- (CA INDEX NAME)



RN 148-24-3 HCPLUS
 CN 8-Quinolinol (CA INDEX NAME)



IC ICM H01L
 CC 37-3 (Plastics Manufacture and Processing)
 Section cross-reference(s): 73
 IT Electroluminescent devices
 (metal 8-hydroxyquinoline-functionalized polymers and their preparation for optical materials)
 IT 97-93-8, Triethylaluminum, reactions 148-24-3,

8-Hydroxyquinoline, reactions 2598-30-3, 5-Formyl-8-hydroxyquinoline
 17016-12-5 511243-83-7, Bicyclo[2.2.1]hept-5-ene-2-hexanenitrile
 (metal 8-hydroxyquinoline-functionalized polymers and their preparation
 for optical materials)

OS.CITING REF COUNT: 3 THERE ARE 3 CAPLUS RECORDS THAT CITE THIS
 RECORD (3 CITINGS)
 REFERENCE COUNT: 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR
 THIS RECORD. ALL CITATIONS AVAILABLE IN THE
 RE FORMAT

L62 ANSWER 24 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2004:638852 HCAPLUS Full-text
 DOCUMENT NUMBER: 141:342388
 TITLE: Investigations on the electronic effects of the
 peripheral 4'-group on
 5-(4'-substituted)phenylazo-8-hydroxyquinoline
 ligands: zinc and aluminium complexes
 La Deda, Massimo; Grisolia, Annarita; Aiello,
 Iolinda; Crispini, Alessandra; Ghedini, Mauro;
 Belviso, Sandra; Amati, Mario; Lelj, Francesco
 LASCAMM, Unita INSTM della Calabria, Dipartimento
 di Chimica, Universita degli Studi della Calabria,
 Arcavacata di Rende), I-87036, Italy
 SOURCE: Dalton Transactions (2004), (16),
 2424-2431
 CODEN: DTARAF; ISSN: 1477-9226
 PUBLISHER: Royal Society of Chemistry
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 OTHER SOURCE(S): CASREACT 141:342388

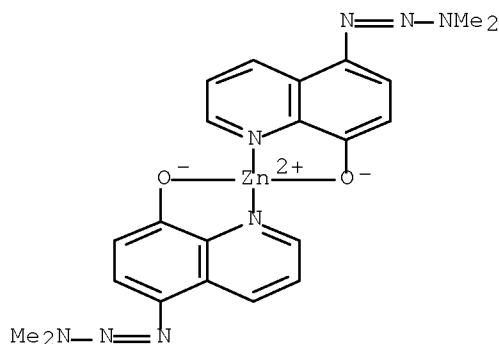
ED Entered STN: 09 Aug 2004

AB 5-(4'-Substituted)phenylazo-8-hydroxyquinolines ($H[L-R]$; $R = NMe_2$, I, C_2H_5 , II, Bu , III, CMe_3 , IV, H , V and F, VI) were prepared and the corresponding $Zn[L-R]_2$ (1a-6a) and $Al[L-R]_3$ (1b-6b) complexes successfully synthesized. These compds. were studied to design new mol. materials with enhanced electron transport properties. The obtained species were extensively characterized by absorption and emission spectra and by cyclic voltammetric measurements. Exptl. and computational results show that the $Zn[L-NMe_2] \cdot 2H_2O$ (1a) and $Al[L-NMe_2]$ (1b) complexes only feature luminescence (at 620 and 600 nm), resp. The unique effects, which are induced by the $N:N-C_6H_4-NMe_2$ group, are further proved by a reversible electron transfer process detected by cyclic voltammetry. These outcomes, discussed from theor. calcns. performed on the $\{H[L-NMe_2]\}^-$, $H[L-NMe_2]$ and $\{H[L-NMe_2]\}^+$ species, suggest that metal complexes formed by 5-(4'-N,N-dimethylamino)phenylazo-8-hydroxyquinoline should be considered as electron transport materials suitable for applications in photonic devices.

IT 769936-04-1
 (cyclic voltammetry and luminescence spectra of)

RN 769936-04-1 HCAPLUS

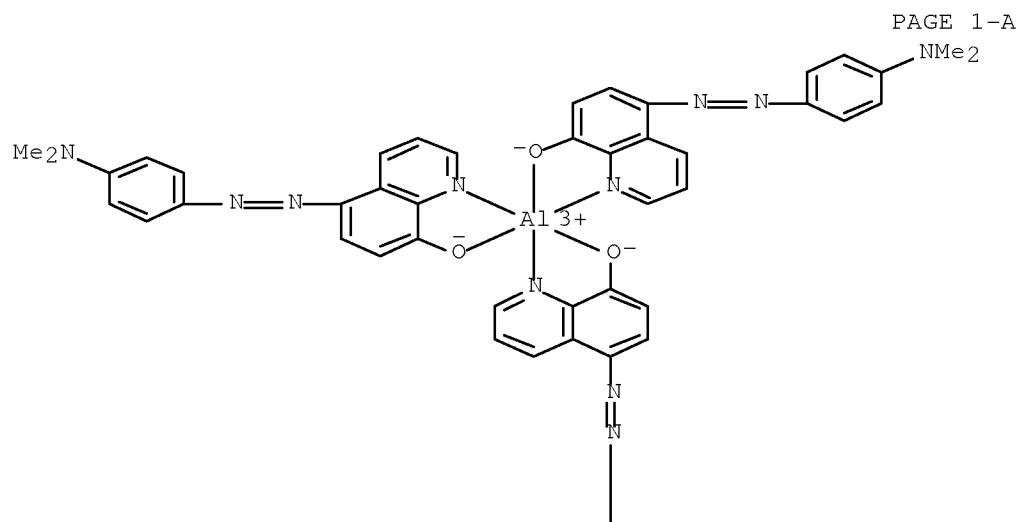
CN Zinc, bis[5-(3,3-dimethyl-1-triazenyl)-8-quinolinolato-
 $\kappa N_1, \kappa O_8]^-$, (T-4)- (9CI) (CA INDEX NAME)



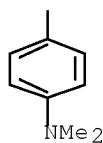
IT 769935-97-9P

(preparation and cyclic voltammetry and fluorescence of)

RN 769935-97-9 HCPLUS

CN Aluminum, tris[5-[[4-(dimethylamino)phenyl]azo]-8-quinolinolato- κ N1, κ O8]- (9CI) (CA INDEX NAME)

PAGE 2-A

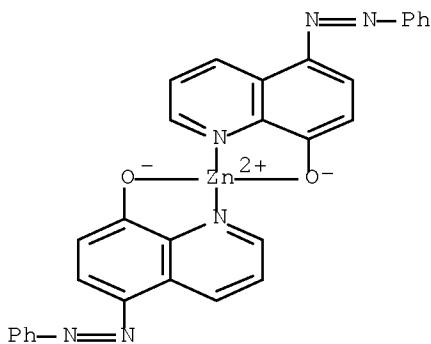
IT 14708-01-1P 769935-93-5P 769935-94-6P
769935-95-7P 769935-96-8P 769935-98-0P

769935-99-1P 769936-00-7P 769936-01-8P
 769936-02-9P

(preparation and cyclic voltammetry of)

RN 14708-01-1 HCAPLUS

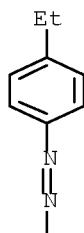
CN Zinc, bis[5-(phenylazo)-8-quinolinolato- κ N1, κ O8]-, (T-4)-
 (9CI) (CA INDEX NAME)



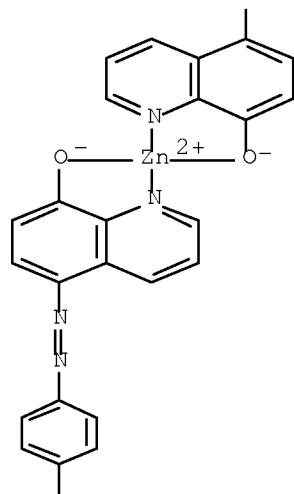
RN 769935-93-5 HCAPLUS

CN Zinc, bis[5-[4-ethylphenyl]azo]-8-quinolinolato- κ N1, κ O8]-
 , (T-4)- (9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 2-A

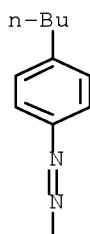


PAGE 3-A

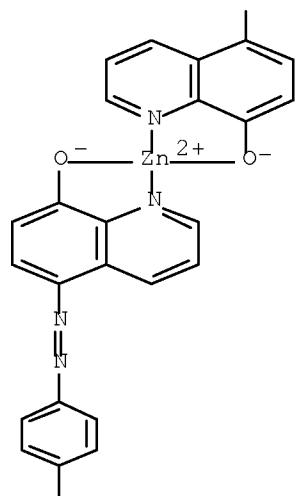
Et

RN 769935-94-6 HCAPLUS
 CN Zinc, bis[5-[(4-butylphenyl)azo]-8-quinolinolato- κ N1, κ O8]-(T-4)- (9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 2-A

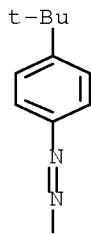


PAGE 3-A

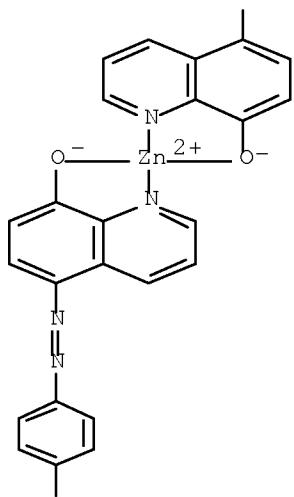
 $n-\text{Bu}^+$

RN 769935-95-7 HCAPLUS
 CN Zinc, bis[5-[4-(1,1-dimethylethyl)phenyl]azo]-8-quinolinolato-
 $\kappa\text{N}1,\kappa\text{O}8]2-, (\text{T}-4)-$ (9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 2-A



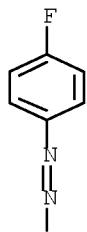
PAGE 3-A

 $t\text{-Bu}$

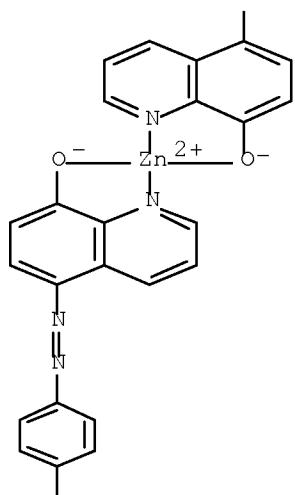
RN 769935-96-8 HCAPLUS

CN Zinc, bis[5-[(4-fluorophenyl)azo]-8-quinolinolato- $\kappa N1,\kappa O8$]-,
(T-4)- (9CI) (CA INDEX NAME)

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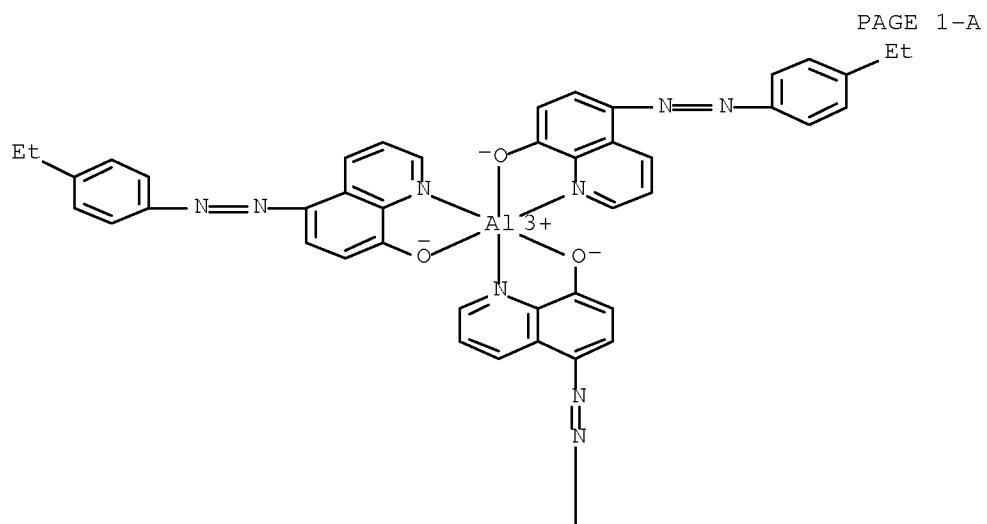
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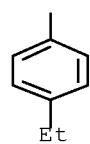
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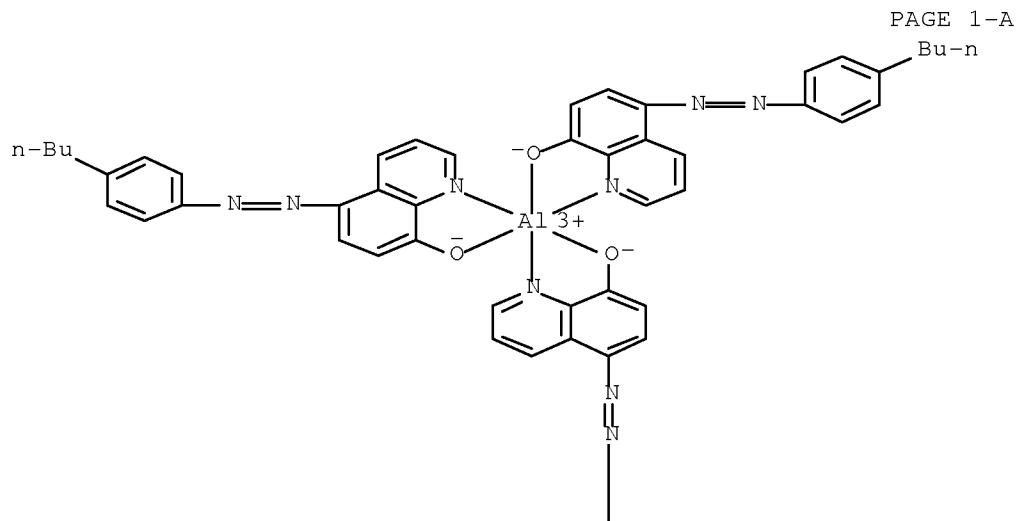
RN 769935-98-0 HCAPLUS
CN Aluminum, tris[5-[(4-ethylphenyl)azo]-8-quinolinolato-
κN1,κO8]- (9CI) (CA INDEX NAME)



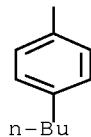
PAGE 2-A



RN 769935-99-1 HCPLUS
 CN Aluminum, tris[5-[(4-butylphenyl)azo]-8-quinolinolato-
 $\kappa\text{N1},\kappa\text{O8}]$ - (9CI) (CA INDEX NAME)

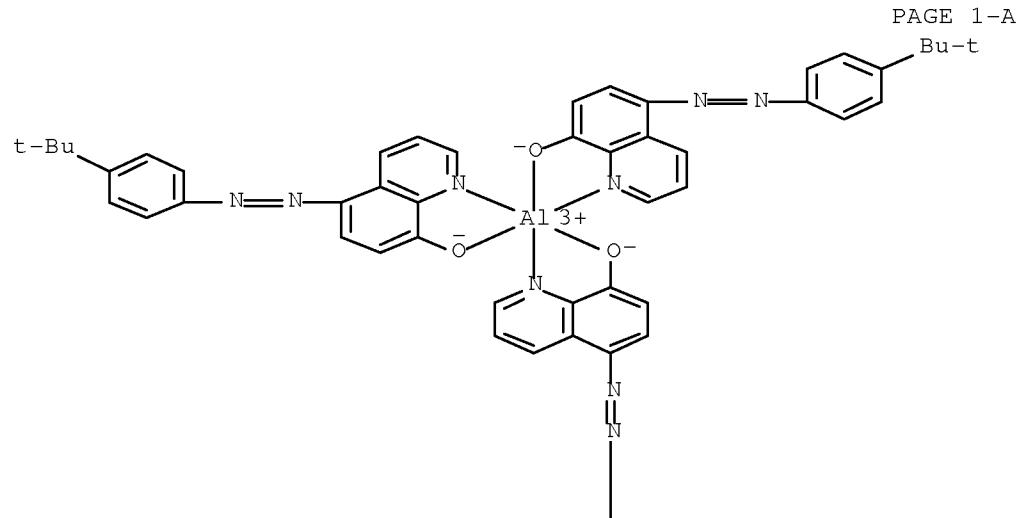


PAGE 2-A

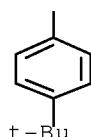


RN 769936-00-7 HCAPLUS
 CN Aluminum, tris[5-[4-(1,1-dimethylethyl)phenyl]azo]-8-quinolinolato-
 κ N1, κ O8]- (9CI) (CA INDEX NAME)

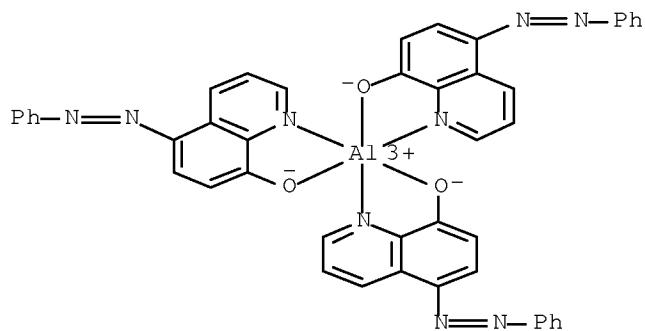
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PAGE 2-A



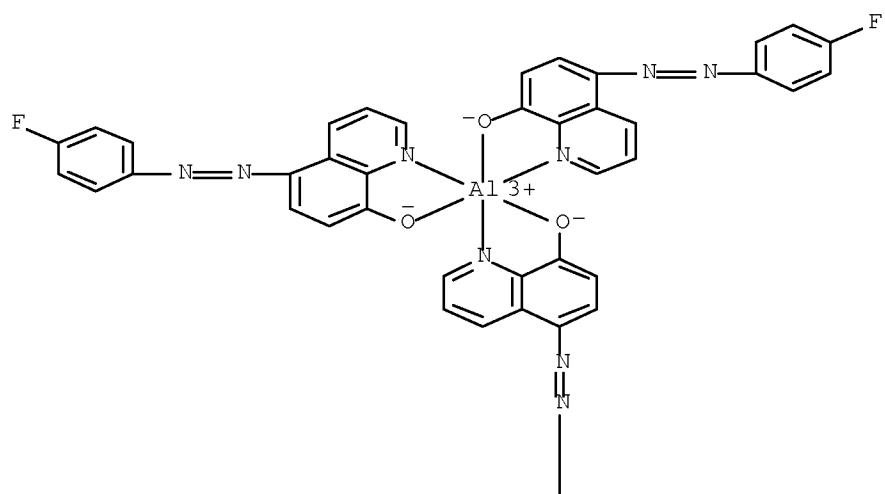
RN 769936-01-8 HCAPLUS
 CN Aluminum, tris[5-(phenylazo)-8-quinolinolato- κ N1, κ O8]- (9CI) (CA INDEX NAME)



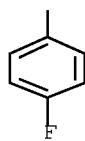
RN 769936-02-9 HCPLUS

CN Aluminum, tris[5-[(4-fluorophenyl)azo]-8-quinolinolato-
κN1,κO8]- (9CI) (CA INDEX NAME)

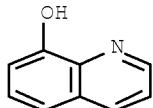
PAGE 1-A



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IT 148-24-3, 8-Hydroxyquinoline, reactions
(reactant for preparation of phenylazohydroxyquinolines)

RN 148-24-3 HCAPLUS
 CN 8-Quinolinol (CA INDEX NAME)



CC 78-7 (Inorganic Chemicals and Reactions)
 Section cross-reference(s): 27, 65, 72, 73, 75
 IT 769936-04-1
 (cyclic voltammetry and luminescence spectra of)
 IT 769935-97-9P
 (preparation and cyclic voltammetry and fluorescence of)
 IT 14708-01-1P 769935-93-5P 769935-94-6P
 769935-95-7P 769935-96-8P 769935-98-0P
 769935-99-1P 769936-00-7P 769936-01-8P
 769936-02-9P
 (preparation and cyclic voltammetry of)
 IT 148-24-3, 8-Hydroxyquinoline, reactions 371-40-4,
 4-Fluoroaniline 769-92-6, 4-tert-Butylaniline
 (reactant for preparation of phenylazohydroxyquinolines)

OS.CITING REF COUNT: 4 THERE ARE 4 CAPLUS RECORDS THAT CITE THIS
 RECORD (4 CITINGS)

REFERENCE COUNT: 48 THERE ARE 48 CITED REFERENCES AVAILABLE FOR
 THIS RECORD. ALL CITATIONS AVAILABLE IN THE
 RE FORMAT

L62 ANSWER 25 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2004:528452 HCAPLUS [Full-text](#)
 DOCUMENT NUMBER: 141:284916
 TITLE: Solution-Processible Small Molecular Organic
 Light-Emitting Diode Material
 and Devices Based on the Substituted Aluminum
 Quinolate
 AUTHOR(S): Cheng, Jung-An; Chen, Chin H.; Liao, Chi Hung
 CORPORATE SOURCE: Department of Applied Chemistry and Display
 Institute, Microelectronics and Information System
 Research Center, National Chiao Tung University,
 Hsinchu, 30050, Taiwan

SOURCE: Chemistry of Materials (2004), 16(15),
 2862-2868
 CODEN: CMATEX; ISSN: 0897-4756

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal

LANGUAGE: English

ED Entered STN: 02 Jul 2004

AB We have discovered a new type of small mol. host material based on aluminum(III) 8-hydroxyquinolates, Al(Saq)3, which was synthesized with three 5-(N-ethylanilinesulfonamide)-8-quinolinolone as bidentate ligands. By X-ray diffraction crystallog. anal., the crystals of meridional Al(Saq)3 are monoclinic, space group P21/c, $a = 17.952(3)$ Å, $b = 17.716(3)$ Å, $c = 17.080(3)$ Å, $\beta = 99.895(4)^\circ$. Its peak photoluminescence in solid phase appears at 488 nm. Its LUMO/HOMO (-3.13/-6.04 eV) and optical band gap (E_g 2.91 eV) were determined by cyclic voltammetry. In solid thin film morphol. investigation,

it shows good thermal properties and high quantum efficiency. When doped with 0.7 weight % of the high fluorescent green dopant 10-(2-benzothiazolyl)-1,1,7,7-tetramethyl-2,3,6,7-tetrahydro-1H,5H,11H-benzo-[1]pyrano[6,7,8-ij]quinolizin-11-one (C-545T), energy transfer from Al(Saq)3 to dopant will occur and high green light emission can be achieved. For fabrication of OLEDs using spin-coating techniques, its electroluminescence is at 1931 CIE_{x,y} (0.21, 0.41).

IT 97-93-8, Triethylaluminum, reactions 148-24-3,

8-Hydroxyquinoline, reactions

(synthesis, structure and optical and thermal properties of Al(Saq)3 electroluminescent material and devices based on Al(Saq)3)

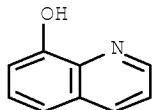
RN 97-93-8 HCPLUS

CN Aluminum, triethyl- (CA INDEX NAME)



RN 148-24-3 HCPLUS

CN 8-Quinolinol (CA INDEX NAME)



CC 73-5 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
Section cross-reference(s): 76, 78

ST electroluminescent device aluminum ethylanilinesulfonamide quinolate photoluminescence band gap structure

IT LUMO (molecular orbital)
(HOMO gap; of Al(Saq)3 electroluminescent material)

IT HOMO (molecular orbital)
(LUMO gap; of Al(Saq)3 electroluminescent material)

IT UV and visible spectra
(absorption; of Al(Saq)3 electroluminescent material)

IT Annealing
(effect on photoluminescence; synthesis, structure and optical and thermal properties of Al(Saq)3 electroluminescent material and devices based on Al(Saq)3)

IT Luminescent substances
(electroluminescent; synthesis, structure, optical and thermal properties of Al(Saq)3 electroluminescent material and devices based on Al(Saq)3)

IT Bond angle

Bond length

Crystal structure

Molecular structure

Surface structure

Thermal stability

X-ray diffraction
 (of Al(Saq)₃ electroluminescent material)

IT Band gap
 (optical; of Al(Saq)₃ electroluminescent material)

IT Electroluminescent devices
 Energy transfer
 Luminescence, electroluminescence
 (synthesis, structure and optical and thermal properties of Al(Saq)₃ electroluminescent material and devices based on Al(Saq)₃)

IT Electric current-potential relationship
 (synthesis, structure, optical and thermal properties of Al(Saq)₃ electroluminescent material and devices based on Al(Saq)₃)

IT Luminescence
 (visible; of Al(Saq)₃ electroluminescent material)

IT 25067-59-8, PVK
 (PVK; synthesis, structure, optical and thermal properties of Al(Saq)₃ electroluminescent material and devices based on Al(Saq)₃)

IT 9003-53-6D, sulfonated
 (dopant for PEDOT; synthesis, structure and optical and thermal properties of Al(Saq)₃ electroluminescent material and devices based on Al(Saq)₃)

IT 126213-51-2
 (doped with PSS; synthesis, structure and optical and thermal properties of Al(Saq)₃ electroluminescent material and devices based on Al(Saq)₃)

IT 155306-71-1, C 545T
 (fluorescent green dopant; synthesis, structure and optical and thermal properties of Al(Saq)₃ electroluminescent material and devices based on Al(Saq)₃)

IT 7429-90-5, Aluminum, uses 7789-24-4, Lithium fluoride LiF, uses 50926-11-9, Indium tin oxide
 (synthesis, structure and optical and thermal properties of Al(Saq)₃ electroluminescent material and devices based on Al(Saq)₃)

IT 757235-20-4P
 (synthesis, structure and optical and thermal properties of Al(Saq)₃ electroluminescent material and devices based on Al(Saq)₃)

IT 97-93-8, Triethylaluminum, reactions 103-69-5,
 N-Ethylaniline 148-24-3, 8-Hydroxyquinoline, reactions 7790-94-5, Chlorosulfonic acid
 (synthesis, structure and optical and thermal properties of Al(Saq)₃ electroluminescent material and devices based on Al(Saq)₃)

IT 757231-25-7P
 (synthesis, structure and optical and thermal properties of Al(Saq)₃ electroluminescent material and devices based on Al(Saq)₃)

IT 123847-85-8
 (synthesis, structure, optical and thermal properties of Al(Saq)₃ electroluminescent material and devices based on Al(Saq)₃)

OS.CITING REF COUNT: 13 THERE ARE 13 CAPLUS RECORDS THAT CITE THIS RECORD (13 CITINGS)

REFERENCE COUNT: 16 THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

ACCESSION NUMBER: 2004:3688 HCAPLUS Full-text
 DOCUMENT NUMBER: 140:67366
 TITLE: AlRq'2 type aluminum compound and method for manufacturing the same and derivative thereof
 INVENTOR(S): Yamamoto, Takakazu; Yamaguchi, Isao; Iijima, Takayuki
 PATENT ASSIGNEE(S): Tokyo Institute of Technology, Japan
 SOURCE: U.S. Pat. Appl. Publ., 14 pp.
 CODEN: USXXCO
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 20040002578	A1	20040101	US 2003-361575 ---<--	20030211
JP 2004026749	A	20040129	JP 2002-187795 ---<--	20020627
PRIORITY APPLN. INFO.:			JP 2002-187795 ---<--	A 20020627

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

OTHER SOURCE(S): MARPAT 140:67366

ED Entered STN: 04 Jan 2004

AB By reacting trialkyl aluminum (AlR₃) with 2-position substituted 8-quinolinol(q'H) at a mole ratio of 1:2, an AlRq'2 type aluminum compound having a specified structural formula is manufactured. The resulting compound is further reacted with an active hydrogen-containing organic compound to obtain a derivative such as an Alq'2q type derivative, wherein q = 8-quinolinolato ligand and q' = 2-position substituted 8-quinolinolato ligand. Thus, 4.60 mL triethylaluminum and 9.55 g 2-methyl-8-quinolinol were reacted at room temperature for 12 h to give 10.9 g ethylbis(2-methyl-8-quinolinolato)aluminum, which could coordinate a conjugated polymer obtained from 5,7-dibromo-8-tert-butyldimethylsilyloxyquinoline and 1,4-diethynyl-2,5-didodecyloxybenzene to give a light emitting macromol. suitable for use as an electroluminescence material.

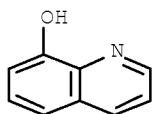
IT 97-93-8, Triethylaluminum, reactions 148-24-3,
 8-Quinolinol, reactions
 (reactant; preparation of alkyldiquinolinol type aluminum compound useful as electroluminescence materials)

RN 97-93-8 HCAPLUS

CN Aluminum, triethyl- (CA INDEX NAME)



RN 148-24-3 HCAPLUS
 CN 8-Quinolinol (CA INDEX NAME)



IC ICM C08G063-78
 ICS C08G063-87; C07F005-06
 INCL 528205000; 546010000
 CC 73-5 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
 Section cross-reference(s): 78
 ST alkyldiquinolinol aluminum compd prepn **electroluminescent** material
 IT Luminescent substances
 (electroluminescent; preparation of alkyldiquinolinol type aluminum compound useful as **electroluminescence** materials)
 IT 535949-58-7P 637772-83-9P 637772-84-0P
 (preparation of alkyldiquinolinol type aluminum compound useful as **electroluminescence** materials)
 IT 469886-06-4P, Aluminum, bis(2-methyl-8-quinolinolato- κ N1, κ O8)(8-quinolinolato- κ N1, κ O8)-, (OC-6-42)-
 (preparation of alkyldiquinolinol type aluminum compound useful as **electroluminescence** materials)
 IT 468758-29-4P, Aluminum, ethylbis(2-methyl-8-quinolinolato- κ N1, κ O8)-
 (preparation of alkyldiquinolinol type aluminum compound useful as **electroluminescence** materials)
 IT 97-93-8, Triethylaluminum, reactions 148-24-3,
 8-Quinolinol, reactions 826-81-3, 2-Methyl-8-quinolinol 2446-69-7,
 4-Hexylphenol
 (reactant; preparation of alkyldiquinolinol type aluminum compound useful as **electroluminescence** materials)

L62 ANSWER 27 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2003:1007243 HCPLUS Full-text
 DOCUMENT NUMBER: 140:51531
 TITLE: Method for the roll-to-roll deposition of an optically transparent and high conductivity metallic thin film
 INVENTOR(S): He, Xiao-Ming; Heydarpour, Ramin
 PATENT ASSIGNEE(S): Avery Dennison Corporation, USA
 SOURCE: PCT Int. Appl., 67 pp.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 2
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2003107079	A2	20031224	WO 2003-US18755	20030612 ---
WO 2003107079	A9	20040304		
WO 2003107079	A3	20040701		
W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD,			

GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ,
 LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ,
 NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ,
 TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW
 RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ,
 BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK,
 EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE,
 SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR,
 NE, SN, TD, TG

US 20040001915	A1	20040101	US 2002-172282	20020614
			<--	
US 6811815	B2	20041102		
AU 2003259035	A1	20031231	AU 2003-259035	20030612
			<--	
EP 1534510	A2	20050601	EP 2003-760348	20030612
			<--	

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC,
 PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK
 PRIORITY APPLN. INFO.: US 2002-172282 A 20020614
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 WO 2003-US18755 W 20030612
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ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

ED Entered STN: 26 Dec 2003

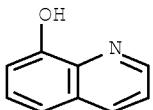
AB The invention relates to a method for the roll-to-roll deposition of an optically transparent and high conductivity metallic thin film, allowing the film to be collected in a continuous roll. The method consists of the steps of (i) providing a flexible plastic substrate; (ii) depositing a multilayered conductive metallic film on the flexible plastic substrate by a thin film deposition technique to form a composite film; and (iii) collecting the composite film in continuous rolls.

IT 148-24-3D, 8-Hydroxyquinoline, metal complexes
 2085-33-8D, Alq₃, derivs.

(luminescent material; method for roll-to-roll deposition
 of optically transparent and high conductivity metallic thin film)

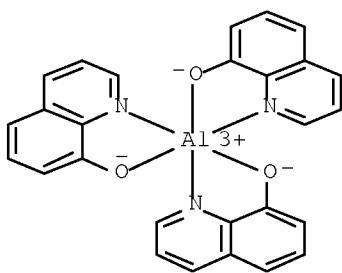
RN 148-24-3 HCPLUS

CN 8-Quinolinol (CA INDEX NAME)



RN 2085-33-8 HCPLUS

CN Aluminum, tris(8-quinolinolato-κN1,κO8)- (CA INDEX NAME)



- IC ICM G02F
 CC 76-2 (Electric Phenomena)
 Section cross-reference(s): 38, 56, 57, 73, 78
 IT Ketones, uses
 (1,3-diketones, metal complexes, luminescent material;
 method for roll-to-roll deposition of optically transparent and
 high conductivity metallic thin film)
 IT Vapor deposition process
 (chemical; method for roll-to-roll deposition of optically transparent
 and high conductivity metallic thin film)
 IT Schiff bases
 (complexes, luminescent material; method for roll-to-roll
 deposition of optically transparent and high conductivity metallic thin
 film)
 IT Vapor deposition process
 (ion plating; method for roll-to-roll deposition of optically
 transparent and high conductivity metallic thin film)
 IT Polyacetylenes, uses
 (ladder, luminescent material; method for roll-to-roll
 deposition of optically transparent and high conductivity metallic thin
 film)
 IT Charge transfer complexes
 Organometallic compounds
 Poly(arylenealkenylenes)
 Polyanilines
 Polypheyls
 Rare earth complexes
 (luminescent material; method for roll-to-roll deposition
 of optically transparent and high conductivity metallic thin film)
 IT Composites
 Laser ablation
 Luminescent substances
 Magnetron sputtering
 (method for roll-to-roll deposition of optically transparent and
 high conductivity metallic thin film)
 IT Ladder polymers
 (polyacetylenes, luminescent material; method for
 roll-to-roll deposition of optically transparent and high conductivity
 metallic thin film)
 IT 91-22-5D, Quinoline, metal complexes 123-54-6D, Acetylacetone, metal
 complexes 148-24-3D, 8-Hydroxyquinoline, metal complexes
 2085-33-8D, Alq₃, derivs. 7440-31-5D, Tin, tin(IV) complexes
 9033-83-4, Poly(phenylene) 17056-99-4D, 5-Hydroxyquinoxaline, metal
 complexes 25038-69-1, Poly(phenylacetylene) 25067-59-8,
 Poly(N-vinylcarbazole) 25233-30-1, Poly(aniline) 25233-34-5,
 Poly(thiophene) 25233-34-5D, Poly(thiophene), 3-alkyl derivs.

26009-24-5, Poly(p-phenylenevinylene) 26009-24-5D,
 Poly(p-phenylenevinylene), dialkoxy derivs. 41999-83-1D,
 Maleonitriledithiol, metal complexes 95270-88-5, Poly(fluorene)
 104934-51-2, Poly(3-octylthiophene)

(luminescent material; method for roll-to-roll deposition
 of optically transparent and high conductivity metallic thin film)

REFERENCE COUNT: 1 THERE ARE 1 CITED REFERENCES AVAILABLE FOR
 THIS RECORD. ALL CITATIONS AVAILABLE IN THE
 RE FORMAT

L62 ANSWER 28 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN

ACCESSION NUMBER: 2003:732139 HCPLUS Full-text

DOCUMENT NUMBER: 139:365902

TITLE: Immiscible polymers in double spin-coated
 electroluminescent devices containing
 phenyl-substituted
 tris(8-hydroxyquinoline)aluminum derivatives
 soluble in a host polymer

AUTHOR(S): Shoji, E.; Miyatake, K.; Hlil, A. R.; Hay, A. S.;
 Maindron, T.; Joussemae, V.; Dodelet, J. P.; Tao,
 Y.; D'Iorio, M.

CORPORATE SOURCE: Department of Chemistry, McGill University,
 Montreal, QC, H3A 2K6, Can.

SOURCE: Journal of Polymer Science, Part A: Polymer
 Chemistry (2003), 41(19), 3006-3016

CODEN: JPACCEC; ISSN: 0887-624X

PUBLISHER: John Wiley & Sons, Inc.

DOCUMENT TYPE: Journal

LANGUAGE: English

ED Entered STN: 18 Sep 2003

AB Three new phenyl-substituted tris(8-hydroxyquinoline)aluminum (AlQ3) derivs.
 have been synthesized:.

Tris(5-phenyl-8-quinolinolate-N1,O8)aluminum, Tris(5,7-diphenyl-8-
 quinolinolate-N1,O8)aluminum, and Tris[5,7-bis(p-fluorophenyl)-8-
 quinolinolate-N1,O8]aluminum. These AlQ3 derivs. are easily soluble in common
 organic solvents and form solid-phase solns. in a poly(aryl ether ketone) host
 polymer (A 435). These interesting properties allow the use of soluble AlQ3
 derivs. in double spin-coated organic light-emitting devices of the type
 ITO/NPB-QP/A 435 + 50 wt% soluble AlQ3 derivative/Mg, where NPB-QP is a hole-
 transporting polymer insol. in toluene, the solvent for A 435. Typical double
 spin-coated organic layer devices are characterized by an emission at 530-539
 nm, a threshold voltage of 6-9 V, and a maximum luminance of 1800-4000 cd/m²
 at 21-25 V.

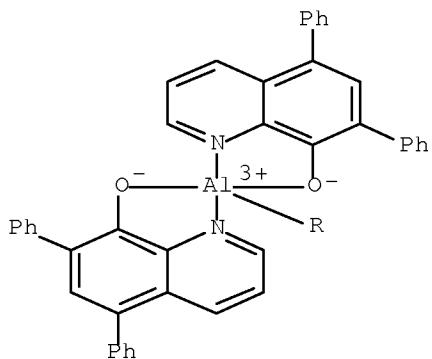
IT 620987-22-6P

(preparation and properties of Ph-substituted
 tris(8-hydroxyquinoline)aluminum derivs. used in double spin-
 coated electroluminescent devices with specially
 designed hole-transporting polymer)

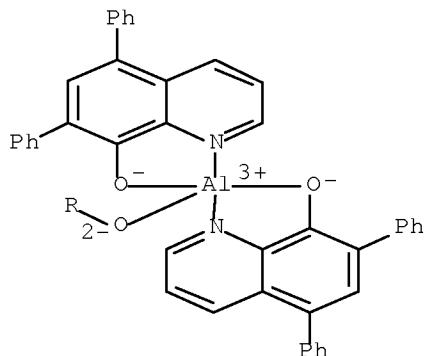
RN 620987-22-6 HCPLUS

CN Aluminum, tetrakis(5,7-diphenyl-8-quinolinolato-κN1,κO8)-
 μ-oxodi- (CA INDEX NAME)

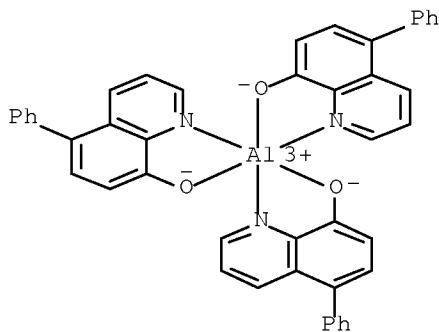
PAGE 1-A



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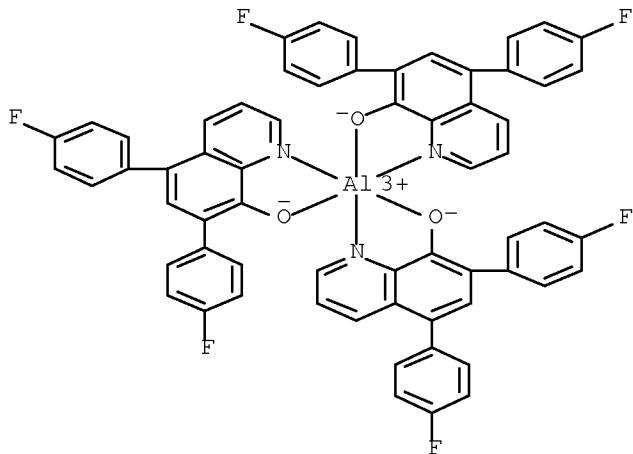


IT 213620-77-0P 269407-70-7P 362623-43-6P
 (preparation of Ph-substituted tris(8-hydroxyquinoline)aluminum derivs.
 used in double spin-coated electroluminescent
 devices with specially designed hole-transporting
 polymer)
 RN 213620-77-0 HCAPLUS
 CN Aluminum, tris(5-phenyl-8-quinolinolato-κN1,κO8)- (CA
 INDEX NAME)



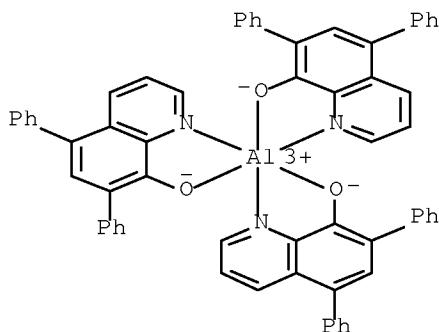
RN 269407-70-7 HCAPLUS

CN Aluminum, tris[5,7-bis(4-fluorophenyl)-8-quinolinolato- κ N1, κ O8]- (CA INDEX NAME)



RN 362623-43-6 HCAPLUS

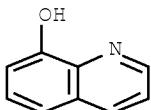
CN Aluminum, tris(5,7-diphenyl-8-quinolinolato- κ N1, κ O8)- (CA INDEX NAME)



IT 97-93-8, Triethylaluminum, reactions 148-24-3,
 8-Hydroxyquinoline, reactions
 (starting material; preparation of Ph-substituted
 tris(8-hydroxyquinoline)aluminum derivs. used in double spin-
 coated electroluminescent devices with specially
 designed hole-transporting polymer)
 RN 97-93-8 HCPLUS
 CN Aluminum, triethyl- (CA INDEX NAME)



RN 148-24-3 HCPLUS
 CN 8-Quinolinol (CA INDEX NAME)



CC 38-3 (Plastics Fabrication and Uses)
 Section cross-reference(s): 29, 73, 76
 ST hydroxyquinoline aluminum phenyl substituted
 electroluminescence host polymer immiscibility; light
 emitting device polyether polyketone host hydroxyquinoline
 aluminum deriv
 IT Polysulfones, uses
 (polyamine-polyether-, aromatic, hole-transporting polymer;
 preparation of Ph-substituted tris(8-hydroxyquinoline)aluminum derivs.
 used in double spin-coated electroluminescent
 devices with specially designed hole-transporting
 polymer)
 IT Polyethers, uses
 (polyamine-polysulfone-, aromatic, hole-transporting
 polymer; preparation of Ph-substituted tris(8-hydroxyquinoline)aluminum
 derivs. used in double spin-coated
 electroluminescent devices with specially designed hole-
 transporting polymer)
 IT Polyketones
 (polyether-, aromatic, host polymer; preparation of Ph-substituted
 tris(8-hydroxyquinoline)aluminum derivs. used in double spin-
 coated electroluminescent devices with specially
 designed hole-transporting polymer)
 IT Polyamines
 (polyether-polysulfone-, aromatic, hole-transporting
 polymer; preparation of Ph-substituted tris(8-hydroxyquinoline)aluminum
 derivs. used in double spin-coated
 electroluminescent devices with specially designed hole-
 transporting polymer)

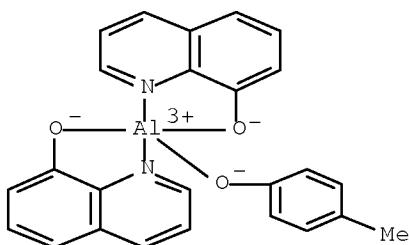
- IT Polyethers, uses
 (polyketone-, aromatic, host polymer; preparation of Ph-substituted tris(8-hydroxyquinoline)aluminum derivs. used in double spin-coated electroluminescent devices with specially designed hole-transporting polymer)
- IT Band gap
 Electric current
 Electroluminescent devices
 Emissivity
 Fluorescence
 Glass transition temperature
 Luminescence, electroluminescence
 Solubility
 Threshold potential
 (preparation and properties of Ph-substituted tris(8-hydroxyquinoline)aluminum derivs. used in double spin-coated electroluminescent devices with specially designed hole-transporting polymer)
- IT 132980-80-4, A 435 (Polymer)
 (A 435, host polymer; preparation of Ph-substituted tris(8-hydroxyquinoline)aluminum derivs. used in double spin-coated electroluminescent devices with specially designed hole-transporting polymer)
- IT 123847-85-8D, NPB, polyether-polysulfones
 (hole-transporting polymer; preparation of Ph-substituted tris(8-hydroxyquinoline)aluminum derivs. used in double spin-coated electroluminescent devices with specially designed hole-transporting polymer)
- IT 1198-14-7P, 5-Bromo-8-hydroxyquinoline 15657-87-1P 84165-50-4P
 202259-06-1P 202259-09-4P 620987-08-8P 620987-09-9P
 620987-10-2P 620987-11-3P
 (intermediate; preparation of Ph-substituted tris(8-hydroxyquinoline)aluminum derivs. used in double spin-coated electroluminescent devices with specially designed hole-transporting polymer)
- IT 620987-22-6P
 (preparation and properties of Ph-substituted tris(8-hydroxyquinoline)aluminum derivs. used in double spin-coated electroluminescent devices with specially designed hole-transporting polymer)
- IT 7439-95-4, Magnesium, uses 50926-11-9, Indium tin oxide
 (preparation of Ph-substituted tris(8-hydroxyquinoline)aluminum derivs. used in double spin-coated electroluminescent devices with specially designed hole-transporting polymer)
- IT 213620-77-0P 269407-70-7P 362623-43-6P
 (preparation of Ph-substituted tris(8-hydroxyquinoline)aluminum derivs. used in double spin-coated electroluminescent devices with specially designed hole-transporting polymer)
- IT 97-93-8, Triethylaluminum, reactions 98-80-6,
 Phenylboronic acid 100-44-7, Benzyl chloride, reactions
 148-24-3, 8-Hydroxyquinoline, reactions 521-74-4
 1765-93-1, p-Fluorophenylboronic acid
 (starting material; preparation of Ph-substituted tris(8-hydroxyquinoline)aluminum derivs. used in double spin-coated electroluminescent devices with specially designed hole-transporting polymer)

OS.CITING REF COUNT: 6 THERE ARE 6 CAPLUS RECORDS THAT CITE THIS RECORD (7 CITINGS)

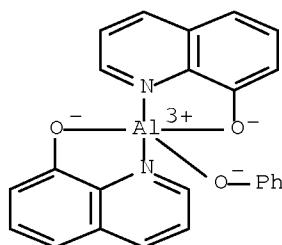
REFERENCE COUNT:

42 THERE ARE 42 CITED REFERENCES AVAILABLE FOR
THIS RECORD. ALL CITATIONS AVAILABLE IN THE
RE FORMAT

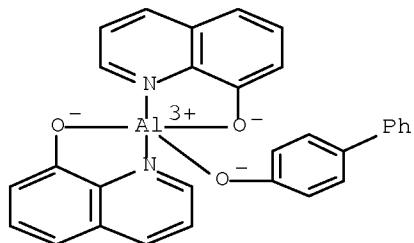
L62 ANSWER 29 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2002:872172 HCPLUS Full-text
 DOCUMENT NUMBER: 138:212400
 TITLE: Novel bis(8-hydroxyquinoline)phenolato-aluminum
 complexes for organic light-emitting diodes
 AUTHOR(S): Wang, Guang; Liang, Fushun; Xie, Zhiyuan; Su,
 Guangping; Wang, Lixiang; Jing, Xiabin; Wang,
 Fosong
 CORPORATE SOURCE: Changchun Institute of Applied Chemistry, The
 State Key Laboratory of Polymer Physics and
 Chemistry, Chinese Academy of Sciences, Changchun,
 130022, Peop. Rep. China
 SOURCE: Synthetic Metals (2002), 131(1-3), 1-5
 CODEN: SYMEDZ; ISSN: 0379-6779
 PUBLISHER: Elsevier Science B.V.
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 ED Entered STN: 18 Nov 2002
 AB Emitting Al complexes containing 2 8-hydroxyquinoline ligands (q) and a
 phenolato ligand (p) were synthesized and characterized. Double layer organic
 light-emitting diodes (OLEDs) were fabricated using these complexes as
 luminescent layers, and strong
 electroluminescence (EL) was observed. Their emitting wavelengths were mainly
 determined by the 1st ligands (q). Cyclic voltammograms revealed a partially
 irreversible n-doping process and indicated that these complexes show
 excellent electron-transporting ability.
 IT 146734-96-5P 224785-36-8P 444716-92-1P
 500298-25-9P
 ((8-hydroxyquinoline)phenolato-aluminum complexes for organic
 LEDs)
 RN 146734-96-5 HCPLUS
 CN Aluminum, (4-methylphenolato)bis(8-quinolinolato-N1,O8)- (9CI) (CA
 INDEX NAME)



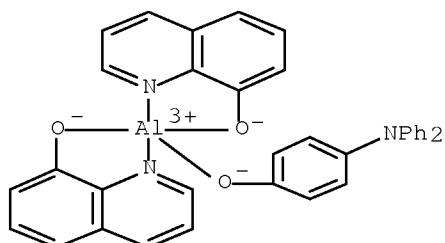
RN 224785-36-8 HCPLUS
 CN Aluminum, phenoxybis(8-quinolinolato- κ N1, κ O8)- (CA INDEX
 NAME)



RN 444716-92-1 HCAPLUS
 CN Aluminum, [[1,1'-biphenyl]-4-olato]bis(8-quinolinolato-
 $\kappa\text{N1}, \kappa\text{O8})-$ (CA INDEX NAME)



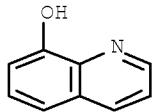
RN 500298-25-9 HCAPLUS
 CN Aluminum, [4-(diphenylamino)phenolato- κO]bis(8-quinolinolato-
 $\kappa\text{N1}, \kappa\text{O8})-$ (CA INDEX NAME)



IT 97-93-8, Triethylaluminum, uses 148-24-3,
 8-Hydroxyquinoline, uses
 ((8-hydroxyquinoline)phenolato-aluminum complexes for organic
 LEDs synthesized using)
 RN 97-93-8 HCAPLUS
 CN Aluminum, triethyl- (CA INDEX NAME)



RN 148-24-3 HCAPLUS
 CN 8-Quinolinol (CA INDEX NAME)



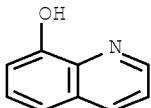
- CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
 Section cross-reference(s): 72, 76, 77
- ST LED hydroxyquinoline phenolato aluminum deriv complex; cyclic voltammetry hydroxyquinoline phenolato aluminum deriv complex LED; electroluminescence hydroxyquinoline phenolato aluminum deriv complex LED; current voltage hydroxyquinoline phenolato aluminum deriv complex LED; NMR spectra hydroxyquinoline phenolato aluminum deriv complex LED; luminescence hydroxyquinoline phenolato aluminum deriv complex LED
- IT Electroluminescent devices ((8-hydroxyquinoline)phenolato-aluminum complexes for organic LEDs)
- IT Cyclic voltammetry Luminescence NMR (nuclear magnetic resonance) (of (8-hydroxyquinoline)phenolato-aluminum complexes for organic LEDs)
- IT Electric current-potential relationship Electric potential Luminescence, electroluminescence (of (8-hydroxyquinoline)phenolato-aluminum complexes in organic LEDs)
- IT 146734-96-3P 224785-36-8P 444716-92-1P 500298-25-9P ((8-hydroxyquinoline)phenolato-aluminum complexes for organic LEDs)
- IT 50926-11-9, ITO 123847-85-8, Npb ((8-hydroxyquinoline)phenolato-aluminum complexes for organic LEDs containing)
- IT 92-69-3, p-Phenylphenol 97-93-8, Triethylaluminum, uses 106-44-5, p-Methylphenol, uses 108-95-2, Phenol, uses 148-24-3, 8-Hydroxyquinoline, uses 25069-86-7, Phenol, 4-diphenylamino- ((8-hydroxyquinoline)phenolato-aluminum complexes for organic LEDs synthesized using)
- OS.CITING REF COUNT: 18 THERE ARE 18 CAPLUS RECORDS THAT CITE THIS RECORD (18 CITINGS)
- REFERENCE COUNT: 10 THERE ARE 10 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L62 ANSWER 30 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2002:626736 HCAPLUS Full-text
 DOCUMENT NUMBER: 137:338348

TITLE: Synthesis of ALQ₃-containing polymers using
 ring-opening metathesis polymerization
 AUTHOR(S): Meyers, Amy; Weck, Marcus
 CORPORATE SOURCE: Sch. Chem. Biochemistry, Georgia Inst. Technology,
 Atlanta, GA, 30332, USA
 SOURCE: Polymer Preprints (American Chemical Society,
 Division of Polymer Chemistry) (2002),
 43(2), 1134
 CODEN: ACPPAY; ISSN: 0032-3934
 PUBLISHER: American Chemical Society, Division of Polymer
 Chemistry
 DOCUMENT TYPE: Journal; (computer optical disk)
 LANGUAGE: English
 ED Entered STN: 20 Aug 2002
 AB A functionalized norbornene unit, containing the attached Alq₃ group, was
 polymerized using ring-opening metathesis. An 8-hydroxyquinoline mol. was
 attached to norbornene; self-assembly of the Alq₂ component gave monomer which
 was then polymerized, resulting in a polymer containing the Alq₃ pendants.
 The UV-visible spectra of both the monomer and polymer show similar absorption
 peaks as Alq₃. Both the monomer and polymer also emitted light at the same
 wavelength (524 nm) as Alq₃. The attachment of the Alq₃-pendant group onto a
 polymer backbone will allow for easy fabrication of electroluminescent
 devices, such as light-emitting diodes.
 IT 97-93-8, Triethylaluminum, reactions 148-24-3,
 8-Hydroxyquinoline, reactions
 (preparation of luminescent
 aluminum-tris(hydroxyquinoline)-pendant polynorbornene by
 ring-opening metathesis polymerization of prepared monomer)
 RN 97-93-8 HCAPLUS
 CN Aluminum, triethyl- (CA INDEX NAME)



RN 148-24-3 HCAPLUS
 CN 8-Quinolinol (CA INDEX NAME)



CC 35-7 (Chemistry of Synthetic High Polymers)
 Section cross-reference(s): 36, 73
 ST hydroxyquinoline norbornene carboxylate monomer prepn ring opening
 polymn; ring opening metathesis polymn Alq₃ pendant polymer prepn;
 luminescence self assembly Alq₃ pendant polymer
 IT Polymerization
 (metathetic, ring-opening; preparation of luminescent
 aluminum-tris(hydroxyquinoline)-pendant polynorbornene by
 ring-opening metathesis polymerization of prepared monomer)

- IT Luminescence
 Self-assembly
 UV and visible spectra
 (preparation of luminescent
 aluminum-tris(hydroxyquinoline)-pendant polynorbornene by
 ring-opening metathesis polymerization of prepared monomer)
- IT 474096-45-2P
 (Alq₃ monomer; preparation of luminescent
 aluminum-tris(hydroxyquinoline)-pendant polynorbornene by
 ring-opening metathesis polymerization of prepared monomer)
- IT 246047-72-3
 (ROMP catalyst; preparation of luminescent
 aluminum-tris(hydroxyquinoline)-pendant polynorbornene by
 ring-opening metathesis polymerization of prepared monomer)
- IT 474096-44-1P
 (intermediate; preparation of luminescent
 aluminum-tris(hydroxyquinoline)-pendant polynorbornene by
 ring-opening metathesis polymerization of prepared monomer)
- IT 474096-46-3P
 (preparation of luminescent
 aluminum-tris(hydroxyquinoline)-pendant polynorbornene by
 ring-opening metathesis polymerization of prepared monomer)
- IT 97-93-8, Triethylaluminum, reactions 148-24-3,
 8-Hydroxyquinoline, reactions 27063-48-5
 (preparation of luminescent
 aluminum-tris(hydroxyquinoline)-pendant polynorbornene by
 ring-opening metathesis polymerization of prepared monomer)

REFERENCE COUNT: 17 THERE ARE 17 CITED REFERENCES AVAILABLE FOR
 THIS RECORD. ALL CITATIONS AVAILABLE IN THE
 RE FORMAT

L62 ANSWER 31 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2002:309105 HCPLUS Full-text
 DOCUMENT NUMBER: 137:207881
 TITLE: Red light-emitting material
 for organic electroluminescence
 AUTHOR(S): Shao, Yan; Liu, Yuhong; Qiu, Yong
 CORPORATE SOURCE: Department of Chemistry, Tsinghua University,
 Beijing, 100084, Peop. Rep. China
 SOURCE: Gongneng Cailiao (2001), 32(6), 662-663,
 666
 CODEN: GOCAEA; ISSN: 1001-9731
 PUBLISHER: Gongneng Cailiao Bianjibu
 DOCUMENT TYPE: Journal
 LANGUAGE: Chinese

ED Entered STN: 25 Apr 2002

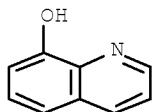
AB The red light-emitting material used in organic electroluminescence, an Al complex, (2,2'-dihydroxyazobenzene)-(8-quinolinoato) Al (Al(azb-q)), was prepared by using AlC₁₃.6H₂O, 2,2'-dihydroxyazobenzene, and 8-hydroxyquinoline. The material had high decomposition temperature of 200° and electron transport ability. The maximum wavelength of the EL emission of Al(azb-q) was at 635 nm. The color position in the color coordinates system showed a strong potential as a red light-emitting material for organic electroluminescence.

IT 148-24-3, 8-Hydroxyquinoline, uses 7784-13-6,
 Aluminum chloride hexahydrate

(in preparation of (2,2'-Dihydroxyazobenzene)(8-quinolinoato)aluminum)

RN 148-24-3 HCPLUS

CN 8-Quinolinol (CA INDEX NAME)

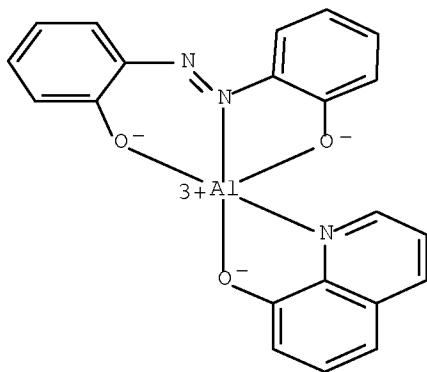


RN 7784-13-6 HCPLUS
 CN Aluminum chloride (AlCl₃), hydrate (1:6) (CA INDEX NAME)



●₆ H₂O

IT 351530-99-9P
 (red light-emitting material for organic
 electroluminescence)
 RN 351530-99-9 HCPLUS
 CN Aluminum, [[2,2'-(azo-κN)-bis[phenolato-κO]](2-)](8-
 quinolinolato-κN1,κO8)- (9CI) (CA INDEX NAME)



CC 73-5 (Optical, Electron, and Mass Spectroscopy and Other
 Related Properties)
 Section cross-reference(s): 74, 78
 ST red light emitting material org
 electroluminescence prepn; hydroxyazobenzene quinolinoato
 aluminum
 IT Luminescence, electroluminescence
 (red; of (2,2'-Dihydroxyazobenzene)(8-quinolinoato)aluminum)
 IT 148-24-3, 8-Hydroxyquinoline, uses 2050-14-8,
 2,2'-Dihydroxyazobenzene 7784-13-6, Aluminum chloride
 hexahydrate
 (in preparation of (2,2'-Dihydroxyazobenzene)(8-quinolinoato)aluminum)

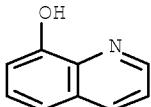
IT 351530-99-9P
 (red light-emitting material for organic
 electroluminescence)

OS.CITING REF COUNT: 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS
 RECORD (1 CITINGS)

L62 ANSWER 32 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2001:376188 HCAPLUS Full-text
 DOCUMENT NUMBER: 135:146235
 TITLE: Synthesis and luminescence behaviors of
 aluminum complex with mixed ligands
 AUTHOR(S): Jang, H.; Do, L.-M.; Kim, Y.; Gon Kim, J.; Zyung,
 T.; Do, Y.
 CORPORATE SOURCE: Department of Chemistry, School of Molecular
 Science, Taejon, 305-600, S. Korea
 SOURCE: Synthetic Metals (2001), 121(1-3),
 1669-1670
 CODEN: SYMEDZ; ISSN: 0379-6779
 PUBLISHER: Elsevier Science S.A.
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 OTHER SOURCE(S): CASREACT 135:146235
 ED Entered STN: 25 May 2001
 AB A novel mixed ligand complex, AlQ(ClQ)₂ (HQ = 8-quinolinol, HClQ = 5,7-dichloro-8-quinolinol) was synthesized and characterized. An organic electroluminescent (EL) device ITO/TPD/AlQ(ClQ)₂/LiF/Al (ITO = In-Sn oxide, TPD = N,N'-diphenyl-N,N'-bis(3-methylphenyl)-1,1'-biphenyl-4,4'-diamine) was employed to study their EL properties. The EL device exhibits green light with maximum luminescence of 780 cd/m² at 6.7 V.
 IT 97-93-8, Triethylaluminum, reactions 148-24-3,
 8-Quinolinol, reactions
 (reactant for preparation of aluminum quinolinolate
 dichloroquinolinolate complex)
 RN 97-93-8 HCAPLUS
 CN Aluminum, triethyl- (CA INDEX NAME)



RN 148-24-3 HCAPLUS
 CN 8-Quinolinol (CA INDEX NAME)



CC 78-7 (Inorganic Chemicals and Reactions)
 ST aluminum quinolinol chloroquinolinol complex prepn
 luminescence

IT Electroluminescent devices
 (green-emitting; using aluminum quinolinolate dichloroquinolinolate complex)
 IT Luminescence
 (of aluminum quinolinolate dichloroquinolinolate complex)
 IT 352004-29-6P
 (preparation and luminescence and electroluminescent device using)
 IT 97-93-8, Triethylaluminum, reactions 148-24-3,
 8-Quinolinol, reactions 773-76-2, 5,7-Dichloro-8-quinolinol
 (reactant for preparation of aluminum quinolinolate dichloroquinolinolate complex)
 OS.CITING REF COUNT: 13 THERE ARE 13 CAPLUS RECORDS THAT CITE THIS RECORD (13 CITINGS)
 REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L62 ANSWER 33 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2001:185142 HCPLUS Full-text
 DOCUMENT NUMBER: 134:246046
 TITLE: Efficient electron-injection for organic electroluminescent devices
 INVENTOR(S): Madathil, Joseph K.; Mason, Max Garrett; Tang, Ching Wan
 PATENT ASSIGNEE(S): Eastman Kodak Company, USA
 SOURCE: Eur. Pat. Appl., 12 pp.
 CODEN: EPXXDW
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 1083612	A2	20010314	EP 2000-202921	20000821 ---
EP 1083612	A3	20040102		
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO				
US 6278236	B1	20010821	US 1999-387402	19990902 ---
JP 2001085165	A	20010330	JP 2000-267679	20000904 ---
PRIORITY APPLN. INFO.:			US 1999-387402	A 19990902 ---

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

ED Entered STN: 16 Mar 2001
 AB An organic electroluminescent (EL) device having a layered structure, including an anode; an organic hole-transport layer in contact with the anode; an organic emitting layer having 1 surface thereof in contact with the hole-transport layer; an organic electron-transport layer in contact with a second surface of the emitting layer; an electron-injecting layer in contact with the electron-transport layer; and a cathode in contact with the electron-injecting layer, in which the electron-injecting layer includes Al and ≥1 alkali halide or ≥1 alkaline earth halide.
 IT 7429-90-5, Aluminum, processes
 (electron-injection layer; efficient electron-injection layers for organic electroluminescent devices)
 RN 7429-90-5 HCPLUS

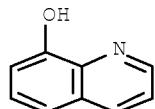
CN Aluminum (CA INDEX NAME)

Al

IT 148-24-3, 8-Quinolinol, processes
 (emitting layer; efficient electron-injection layers for organic
 electroluminescent devices)

RN 148-24-3 HCPLUS

CN 8-Quinolinol (CA INDEX NAME)



IT 37275-76-6, Aluminum Zinc oxide 117944-65-7,
 Indium Zinc oxide
 (light transmissive anode; efficient electron-injection layers for
 organic electroluminescent devices)

RN 37275-76-6 HCPLUS

CN Aluminum zinc oxide (CA INDEX NAME)

Component	Ratio	Component	
			Registry Number
O	x		17778-80-2
Zn	x		7440-66-6
Al	x		7429-90-5

RN 117944-65-7 HCPLUS

CN Indium zinc oxide (CA INDEX NAME)

Component	Ratio	Component	
			Registry Number
O	x		17778-80-2
In	x		7440-74-6
Zn	x		7440-66-6

IC ICM H01L051-20

CC 76-3 (Electric Phenomena)
 Section cross-reference(s): 74, 75

ST efficient electron injection org electroluminescent device

IT Electron transport
 Electron-hole recombination
 Hole transport
 (efficient electron-injection layers for organic
 electroluminescent devices)

IT Transparent films
 (elec. conductive, anode; efficient electron-injection layers for
 organic electroluminescent devices)

IT Alkali metal fluorides
 Alkaline earth fluorides
 (electron-injection layer; efficient electron-injection layers for organic electroluminescent devices)

IT Conduction electrons
 (electron-injection; efficient electron-injection layers for organic electroluminescent devices)

IT Electric conductors
 (films, transparent, anode; efficient electron-injection layers for organic electroluminescent devices)

IT Electroluminescent devices
 (organic; efficient electron-injection layers for organic electroluminescent devices)

IT Glass, uses
 Plastics, uses
 (substrate; efficient electron-injection layers for organic electroluminescent devices)

IT Vapor deposition process
 (vacuum; efficient electron-injection layers for organic electroluminescent devices)

IT 7429-90-5, Aluminum, processes
 (electron-injection layer; efficient electron-injection layers for organic electroluminescent devices)

IT 7681-49-4, Sodium fluoride, processes 7783-40-6, Magnesium fluoride
 7783-48-4, Strontium fluoride 7787-32-8, Barium fluoride
 7789-23-3, Potassium fluoride 7789-24-4, Lithium fluoride, processes
 7789-75-5, Calcium fluoride, processes 13400-13-0, Cesium fluoride
 13446-74-7, Rubidium fluoride
 (electron-injection layers; efficient electron-injection layers for organic electroluminescent devices)

IT 148-24-3, 8-Quinolinol, processes
 (emitting layer; efficient electron-injection layers for organic electroluminescent devices)

IT 1332-29-2, Tin oxide 12640-79-8, Nickel tungsten oxide
 37275-76-6, Aluminum Zinc oxide 50926-11-9, Indium-tin-oxide
 56997-34-3, Cadmium tin oxide 117944-65-7, Indium Zinc oxide 158346-28-2, Indium Magnesium oxide
 (light transmissive anode; efficient electron-injection layers for organic electroluminescent devices)

IT 14808-60-7, Quartz, uses
 (substrate; efficient electron-injection layers for organic electroluminescent devices)

OS.CITING REF COUNT: 6 THERE ARE 6 CAPLUS RECORDS THAT CITE THIS RECORD (6 CITINGS)

REFERENCE COUNT: 1 THERE ARE 1 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L62 ANSWER 34 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 2001:6947 HCAPLUS Full-text
 DOCUMENT NUMBER: 134:185686
 TITLE: Material transport regimes and mechanisms for growth of molecular organic thin films using low-pressure organic vapor phase deposition
 AUTHOR(S): Shtein, Max; Gossenberger, Herman F.; Benziger, Jay B.; Forrest, Stephen R.
 CORPORATE SOURCE: Center for Photonics and Optoelectronic Materials and Department of Chemical Engineering, Princeton University, Princeton, NJ, 08544, USA
 SOURCE: Journal of Applied Physics (2001),

89(2), 1470-1476

CODEN: JAPIAU; ISSN: 0021-8979

PUBLISHER:

American Institute of Physics

DOCUMENT TYPE:

Journal

LANGUAGE:

English

ED Entered STN: 04 Jan 2001

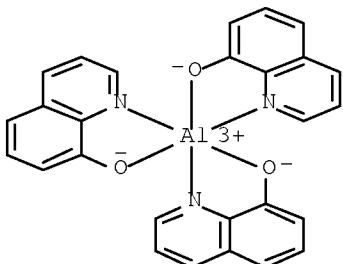
AB The authors determine the phys. mechanisms controlling the growth of amorphous organic thin films by the process of low-pressure organic vapor phase deposition (LP-OVPD). In LP-OVPD, multiple host and dopant mol. sources are introduced into a hot wall reactor via several injection barrels using an inert carrier gas, allowing for controlled film growth rates exceeding 10 Å/s. The temperature and carrier flow rate for each source can be independently regulated, allowing considerable control over dopant concentration, deposition rate, and thickness uniformity of the thin films. The rate of film deposition is limited either by the rate of condensation on the substrate or by the rate of supply from the source. The source-limited regime can be further classified into equilibrium or kinetically limited evaporation, coupled to convection- or diffusion-limited deposition. Models are developed to relate the rate of film growth to source and substrate temperature, and carrier gas flow rate. These models characterize and predict the performance of the LP-OVPD system used to grow high performance organic light emitting devices.

IT 2085-33-8, Tris(8-hydroxyquinoline)aluminum

(material transport regimes and mechanisms for growth of mol. organic thin films using low-pressure organic vapor phase deposition)

RN 2085-33-8 HCPLUS

CN Aluminum, tris(8-quinolinolato-κN1,κO8)- (CA INDEX NAME)



CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
Section cross-reference(s): 74, 75, 76

IT Convective flow

Dopants

Electroluminescent devices

Flow

Fluorescent dyes

Simulation and Modeling, physicochemical

Thickness

Vapor deposition process

(material transport regimes and mechanisms for growth of mol. organic thin films using low-pressure organic vapor phase deposition)

IT 147-14-8, Copper phthalocyanine 2085-33-8,

Tris(8-hydroxyquinoline)aluminum 31248-39-2, Platinum

octaethylporphyrin 58328-31-7, 4,4'-N,N'-Dicarbazolylbiphenyl

94928-86-6 123847-85-8, α-NPD

(material transport regimes and mechanisms for growth of mol. organic

thin films using low-pressure organic vapor phase deposition)
OS.CITING REF COUNT: 51 THERE ARE 51 CAPLUS RECORDS THAT CITE THIS
RECORD (51 CITINGS)
REFERENCE COUNT: 20 THERE ARE 20 CITED REFERENCES AVAILABLE FOR
THIS RECORD. ALL CITATIONS AVAILABLE IN THE
RE FORMAT

L62 ANSWER 35 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN
ACCESSION NUMBER: 2000:536215 HCPLUS [Full-text](#)
DOCUMENT NUMBER: 133:238413
TITLE: Synthesis and characterization of a novel
AlQ₃-containing polymer
AUTHOR(S): Lu, Jianping; Hlil, Antisar R.; Meng, Yuezhong;
Hay, Allan S.; Tao, Ye; D'Iorio, Marie; Maindron,
Tony; Dodelet, Jean-Pol
CORPORATE SOURCE: Department of Chemistry, McGill University,
Montreal, QC, H3A 2K6, Can.
SOURCE: Journal of Polymer Science, Part A: Polymer
Chemistry (2000), 38(16), 2887-2892
CODEN: JPACEC; ISSN: 0887-624X
PUBLISHER: John Wiley & Sons, Inc.
DOCUMENT TYPE: Journal
LANGUAGE: English

ED Entered STN: 06 Aug 2000

AB The synthesis of a tris(8-hydroxyquinoline)aluminum (AlQ₃)-containing poly(arylene ether) (I) is reported. The presence of AlQ₃ pendants in polymer I is confirmed by NMR, UV-visible, photoluminescence, and gel permeation chromatog. analyses. This is the first report of the attachment of AlQ₃ complexes as side chains to a polymer. Polymer I has a glass-transition temperature of 217.8° and is thermally stable with a 5% weight-loss temperature greater than 500° under nitrogen, as determined by differential scanning calorimetry and thermogravimetric analyses, resp. Polymer I is quite soluble in common organic solvents, such as THF, N,N-dimethylacetamide, and CHCl₃. A composite that is 80 wt % polymer I and 20 wt% AlQ₃ forms a transparent and tough film when cast from its chloroform solution. The application of this AlQ₃-containing polymer in light-emitting diodes is under investigation.

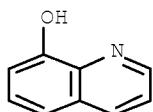
IT 97-93-8DP, Triethylaluminum, reaction products with
poly(arylene ether), aminohydroquinoline, and hydroxyquinoline
148-24-3DP, 8-Hydroxyquinoline, reaction products with
poly(arylene ether), aminohydroquinoline, and triethylaluminum
(synthesis and characterization of novel
tris(8-hydroxyquinoline)aluminum-containing poly(arylene ether))

RN 97-93-8 HCPLUS

CN Aluminum, triethyl- (CA INDEX NAME)



RN 148-24-3 HCPLUS
CN 8-Quinolinol (CA INDEX NAME)



CC 35-5 (Chemistry of Synthetic High Polymers)
 Section cross-reference(s): 73, 76
 ST hydroxyquinoline aluminum polyarylene polyether prepn property;
 light emitting diode hydroxyquinoline aluminum
 polyarylene polyether
 IT Electroluminescent devices
 Glass transition temperature
 Luminescence
 Polymerization
 Solubility
 (synthesis and characterization of novel
 tris(8-hydroxyquinoline)aluminum-containing poly(arylene ether))
 IT 97-93-8DP, Triethylaluminum, reaction products with
 poly(arylene ether), aminohydroquinoline, and hydroxyquinoline
 148-24-3DP, 8-Hydroxyquinoline, reaction products with
 poly(arylene ether), aminohydroquinoline, and triethylaluminum
 21302-43-2DP, 5-Amino-8-hydroxyquinoline dihydrochloride, reaction
 products with poly(arylene ether), complexes with aluminum and
 hydroquinoline 294212-77-4DP, reaction products with
 aminohydroquinoline, complexes with aluminum and hydroquinoline
 (synthesis and characterization of novel
 tris(8-hydroxyquinoline)aluminum-containing poly(arylene ether))
 OS.CITING REF COUNT: 40 THERE ARE 40 CAPLUS RECORDS THAT CITE THIS
 RECORD (40 CITINGS)
 REFERENCE COUNT: 15 THERE ARE 15 CITED REFERENCES AVAILABLE FOR
 THIS RECORD. ALL CITATIONS AVAILABLE IN THE
 RE FORMAT

L62 ANSWER 36 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 1999:350782 HCAPLUS Full-text
 DOCUMENT NUMBER: 130:359585
 TITLE: Low pressure vapor phase deposition of organic
 thin films
 INVENTOR(S): Forrest, Stephen R.; Burrows, Paul; Ban, Vladimir
 S.
 PATENT ASSIGNEE(S): The Trustees of Princeton University, USA
 SOURCE: PCT Int. Appl., 42 pp.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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WO 9925894	A1	19990527	WO 1998-US24424	19981116

W: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ,
 DE, DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IS,
 JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG,
 MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK,
 SL, TJ, TM, TR, TT, UA, UG, UZ, VN, YU, ZW
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RW: GH, GM, KE, LS, MW, SD, SZ, UG, ZW, AT, BE, CH, CY, DE, DK,
 ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF,
 CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
 US 6337102 B1 20020108 US 1997-972156 19971117
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 AU 9914124 A 19990607 AU 1999-14124 19981116
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 EP 1032722 A1 20000906 EP 1998-957997 19981116
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 EP 1032722 B1 20041027
 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC,
 PT, IE, SI, LT, LV, FI, RO
 JP 2001523768 T 20011127 JP 2000-521253 19981116
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 TW 575699 B 20040211 TW 1998-87118943 19981117
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 US 20010002279 A1 20010531 US 2000-736090 20001213
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 US 20020155230 A1 20021024 US 2002-125400 20020419
 <--
 US 6558736 B2 20030506
 US 20040007178 A1 20040115 US 2003-427933 20030502
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 US 20070131172 A1 20070614 US 2007-655258 20070119
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 JP 2009238756 A 20091015 JP 2009-138382 20090609
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 PRIORITY APPLN. INFO.: US 1997-972156 A 19971117
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 JP 2000-521253 A3 19981116
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 WO 1998-US24424 W 19981116
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 US 2000-663143 B1 20000915
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 US 2000-736090 A1 20001213
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 US 2002-125400 A3 20020419
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 US 2003-427933 A1 20030502
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ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

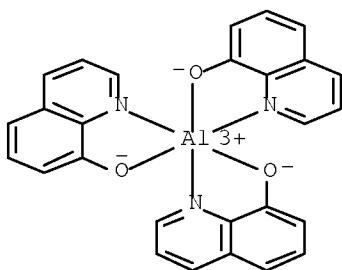
ED Entered STN: 08 Jun 1999

AB Methods for preparing organic thin films on substrates are described which entail providing a plurality of organic precursors in the vapor phase, and reacting the plurality of organic precursors at a pressure below atmospheric pressure to produce a film on the substrate. The methods may be applied to the production of organic light-emitting devices. Apparatus for carrying out the methods is described which comprises a reaction chamber; means for heating the reaction chamber; means for introducing vapors of organic precursor materials into the reaction chamber; and means for reducing the pressure in the reaction chamber to below atmospheric pressure. Apparatus is also described which includes a plurality of vacuum chambers and a conveyor for moving substrates between them. Films, including light-emitting and nonlinear optical material films, formed by the methods are also claimed.

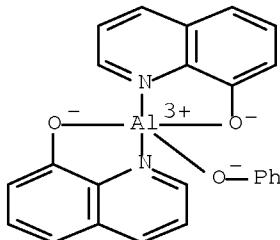
IT 2085-33-8P, Tris(8-hydroxyquinolinato)aluminum
 224785-36-8P

(methods for low pressure vapor phase deposition of organic thin films and deposition apparatus and films produced by the methods)

RN 2085-33-8 HCPLUS

CN Aluminum, tris(8-quinolinolato- κ N1, κ O8)- (CA INDEX NAME)

RN 224785-36-8 HCAPLUS

CN Aluminum, phenoxybis(8-quinolinolato- κ N1, κ O8)- (CA INDEX NAME)

IC ICM C23C016-00

ICS H01J001-62

CC 75-1 (Crystallography and Liquid Crystals)

Section cross-reference(s): 73, 76

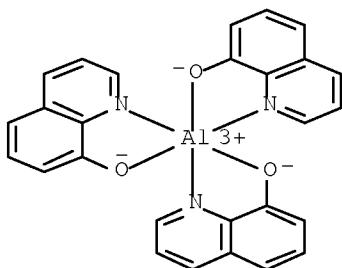
ST low pressure org vapor deposition app; electroluminescent device fabrication low pressure org vapor deposition; nonlinear optical film low pressure org vapor deposition; light emitting film low pressure org vapor deposition

IT Phosphors
(electroluminescent; methods and apparatus for low pressure vapor phase deposition of organic)IT Electroluminescent devices
Electroluminescent devices
Semiconductor device fabrication
(methods and. apparatus for low pressure vapor phase deposition of organic thin films for)IT 917-23-7P, 5,10,15,20-Tetraphenyl-21H,23H-porphine
20885-33-8P, Tris(8-hydroxyquinolinato)aluminum 51325-91-8P,
4-(Dicyanomethylene)-2-methyl-6-(p-dimethylaminostyryl)-4H-pyran
65181-78-4P, N,N'-Diphenyl-N,N'-bis(3-methylphenyl)-1,1'-biphenyl-4,4'-diamine 123847-85-8P 124729-98-2P, MTDATA 224785-36-8P
(methods for low pressure vapor phase deposition of organic thin films and deposition apparatus and films produced by the methods)

OS.CITING REF COUNT: 65 THERE ARE 65 CAPLUS RECORDS THAT CITE THIS RECORD (71 CITINGS)

REFERENCE COUNT: 9 THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L62 ANSWER 37 OF 38 HCPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 1997:756184 HCPLUS Full-text
 DOCUMENT NUMBER: 128:94828
 ORIGINAL REFERENCE NO.: 128:18425a,18428a
 TITLE: Low pressure organic vapor phase deposition of small molecular weight organic light emitting device structures
 AUTHOR(S): Baldo, M. A.; Kozlov, V. G.; Burrows, P. E.; Forrest, S. R.; Ban, V. S.; Koene, B.; Thompson, M. E.
 CORPORATE SOURCE: Princeton Materials Institute, Center for Photonics and Optoelectronic Materials, Department of Electrical Engineering, Princeton University, Princeton, NJ, 08544, USA
 SOURCE: Applied Physics Letters (1997), 71(21), 3033-3035
 CODEN: APPLAB; ISSN: 0003-6951
 PUBLISHER: American Institute of Physics
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 ED Entered STN: 04 Dec 1997
 AB A new technique for the deposition of amorphous organic thin films, low pressure organic vapor phase deposition (LP-OVPD), was used to fabricate organic light emitting devices (OLEDs) consisting of a film of aluminum tris-(8 hydroxyquinoline) (Alq₃) grown on the surface of a film of N'-diphenyl-N,N'-bis(3-methylphenyl)1-1'biphenyl-4-4'diamine. The resulting heterojunction OLED has a performance similar to conventional, small mol. weight OLEDs grown using thermal evaporation in vacuum. The LP-OVPD grown device has an external quantum efficiency of 0.40 ± 0.05 and a turn-on voltage of .apprx.6 V. The rapid throughput demonstrated with LP-OVPD has the potential to facilitate low cost mass production of conventional small mol. based OLEDs, and its use of low vacuum in a horizontal reactor lends itself to roll-to-roll deposition of organic films for many photonic device applications. The OLEDs were grown on both glass and flexible polyester substrates precoated with transparent layers of indium tin oxide (ITO). The cathode material was Mg:Ag alloy (25:1).
 IT 2085-33-8, Tris(8-quinolinolato)aluminum
 (organic light emitting device using low pressure organic vapor phase deposition)
 RN 2085-33-8 HCPLUS
 CN Aluminum, tris(8-quinolinolato-κN1,κO8)- (CA INDEX NAME)



CC 73-5 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
 Section cross-reference(s): 76
 ST electroluminescent device org vapor phase deposition;
 aluminum quinolinolato low pressure vapor deposition
 IT Surface roughness
 (of organic light emitting device by low pressure
 organic vapor phase deposition using AFM)
 IT Electric current-potential relationship
 (of organic light emitting device using low
 pressure organic vapor phase deposition)
 IT Vapor phase epitaxy
 (organic light emitting device by low pressure
 organic vapor phase deposition)
 IT Electroluminescent devices
 (organic light emitting device using low pressure
 organic vapor phase deposition)
 IT Polyesters, uses
 (substrate; organic light emitting device using
 low pressure organic vapor phase deposition)
 IT 2085-33-8, Tris(8-quinolinolato)aluminum 65181-78-4, TPD
 (organic light emitting device using low pressure
 organic vapor phase deposition)

OS.CITING REF COUNT: 28 THERE ARE 28 CAPLUS RECORDS THAT CITE THIS RECORD (28 CITINGS)
 REFERENCE COUNT: 8 THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L62 ANSWER 38 OF 38 HCAPLUS COPYRIGHT 2010 ACS on STN
 ACCESSION NUMBER: 1995:547736 HCAPLUS Full-text
 DOCUMENT NUMBER: 123:69921
 ORIGINAL REFERENCE NO.: 123:12237a,12240a
 TITLE: Manufacture of organic electroluminescent device
 INVENTOR(S): Sato, Yoshiharu; Kanai, Hiroyuki
 PATENT ASSIGNEE(S): Mitsubishi Kagaku KK, Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 8 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

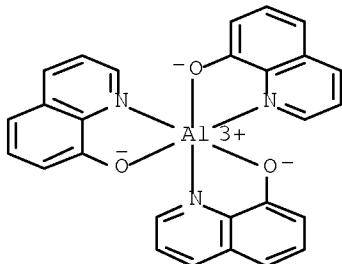
PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 07062526	A	19950307	JP 1993-205375 -->	19930819
PRIORITY APPLN. INFO.:			JP 1993-205375 -->	19930819

ED Entered STN: 13 May 1995
 AB The device is manufactured by forming the organic light-emitting layer on a substrate at 60-150°. The layer may contain a metal complex of 8-hydroxyquinoline, which may be formed by vacuum vapor deposition. The device shows good heat resistance and emission stability for long periods.
 IT 2085-33-8, Tris(8-hydroxyquinolinato)aluminum
 7069-05-8 13978-85-3,
 Bis(8-hydroxyquinolinato)zinc 14642-34-3,
 Tris(8-hydroxyquinolinato)gallium

(manufacture of electroluminescent device containing hydroxyquinoline metal complex)

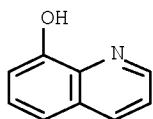
RN 2085-33-8 HCPLUS

CN Aluminum, tris(8-quinolinolato- κ N1, κ O8)- (CA INDEX NAME)



RN 7069-05-8 HCPLUS

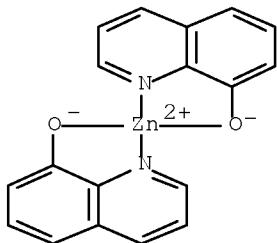
CN 8-Quinolinol, calcium salt (2:1) (CA INDEX NAME)



●1/2 Ca

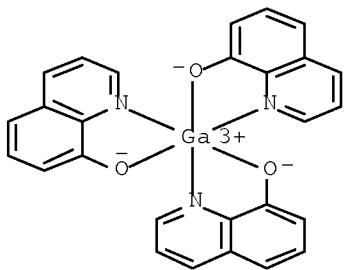
RN 13978-85-3 HCPLUS

CN Zinc, bis(8-quinolinolato- κ N1, κ O8)-, (T-4)- (CA INDEX NAME)



RN 14642-34-3 HCPLUS

CN Gallium, tris(8-quinolinolato- κ N1, κ O8)- (9CI) (CA INDEX NAME)



IC ICM C23C014-24

ICS C09K011-06; H05B033-10; H05B033-14

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

ST electroluminescent device hydroxyquinoline metal complex;
deposition vacuum vapor electroluminescent film

IT Electroluminescent devices

Vapor deposition processes

(manufacture of electroluminescent device containing
hydroxyquinoline metal complex)

IT 2085-33-8, Tris(8-hydroxyquinolinato)aluminum
7069-05-8 13978-85-3,

Bis(8-hydroxyquinolinato)zinc 14514-42-2,

Tris(8-hydroxyquinolinato)indium 14642-34-3,

Tris(8-hydroxyquinolinato)gallium 15276-55-8 15956-38-4,

Tris(8-hydroxyquinolinato)scandium 16009-78-2,

Tris(8-hydroxyquinolinato)yttrium 67952-28-7,

Bis(8-hydroxyquinolinato)magnesium

(manufacture of electroluminescent device containing
hydroxyquinoline metal complex)

OS.CITING REF COUNT: 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS
RECORD (1 CITINGS)

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(FILE 'HOME' ENTERED AT 07:40:40 ON 02 MAR 2010)

FILE 'HCAPLUS' ENTERED AT 07:40:51 ON 02 MAR 2010

L1 1 SEA SPE=ON ABB=ON PLU=ON US20070190247/PN
SEL RN

FILE 'REGISTRY' ENTERED AT 07:41:03 ON 02 MAR 2010

L2 20 SEA SPE=ON ABB=ON PLU=ON (1333-74-0/B1 OR 15318-08-8/B1
OR 19553-62-9/B1 OR 20791-15-5/B1 OR 310888-77-8/B1 OR
310888-80-3/B1 OR 310888-81-4/B1 OR 310888-82-5/B1 OR
310888-85-8/B1 OR 310888-87-0/B1 OR 7358-26-1/B1 OR
7440-37-1/B1 OR 7440-59-7/B1 OR 75-24-1/B1 OR 7727-37-9/B1
OR 870126-56-0/B1 OR 870126-57-1/B1 OR 870126-58-2/B1 OR
870126-59-3/B1 OR 97-93-8/B1)
E 8-HYDROXYQUINOLINATE/CN

L3 1 SEA SPE=ON ABB=ON PLU=ON 8-HYDROXYQUINOLINATE/CN
E 8-HYDROXYQUINOLINE/CN

L4 1 SEA SPE=ON ABB=ON PLU=ON 8-HYDROXYQUINOLINE/CN

L5 685 SEA SPE=ON ABB=ON PLU=ON 148-24-3/CRN

L6 7 SEA SPE=ON ABB=ON PLU=ON L5 AND (AL OR GA OR ZN)/ELS

L7 16 SEA SPE=ON ABB=ON PLU=ON L2 AND M/ELS

L8 663827 SEA SPE=ON ABB=ON PLU=ON (AL OR GA OR ZN)/ELS

L9 122525 SEA SPE=ON ABB=ON PLU=ON L8 AND CCS/CI

L10 541302 SEA SPE=ON ABB=ON PLU=ON L8 NOT L9

L11 541302 SEA SPE=ON ABB=ON PLU=ON L10 OR L10
D 300000 RN

L12 300000 SEA RAN=(173351-91-2) SPE=ON ABB=ON PLU=ON L10 OR L10

L13 241302 SEA SPE=ON ABB=ON PLU=ON L11 NOT L12

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L14 23243 SEA SPE=ON ABB=ON PLU=ON L7

L15 143187 SEA SPE=ON ABB=ON PLU=ON L9

L16 140299 SEA SPE=ON ABB=ON PLU=ON L12

L17 2254991 SEA SPE=ON ABB=ON PLU=ON L13

FILE 'REGISTRY' ENTERED AT 08:23:38 ON 02 MAR 2010

L18 33 SEA SPE=ON ABB=ON PLU=ON L8 AND HYDROXYQUINOL?

L19 9 SEA SPE=ON ABB=ON PLU=ON 16582-16-4/CRN

L20 1 SEA SPE=ON ABB=ON PLU=ON L19 AND (AL OR GA OR ZN)/ELS

FILE 'HCAPLUS' ENTERED AT 08:24:42 ON 02 MAR 2010

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L22 10121 SEA SPE=ON ABB=ON PLU=ON L4

L23 1836 SEA SPE=ON ABB=ON PLU=ON L5

L24 9 SEA SPE=ON ABB=ON PLU=ON L19

L25 6 SEA SPE=ON ABB=ON PLU=ON L6

L26 9679 SEA SPE=ON ABB=ON PLU=ON L18

L27 1 SEA SPE=ON ABB=ON PLU=ON L20

L28 11763 SEA SPE=ON ABB=ON PLU=ON (L14 OR L15 OR L16 OR L17) AND
(L21 OR L22 OR L23 OR L24 OR L25 OR L26 OR L27)

L29 QUE SPE=ON ABB=ON PLU=ON LUM!N? OR ELECTROLUM!N? OR
ORGANOLUM!N? OR (ELECTRO OR ORGANO OR ORG#) (2A)LUM!N? OR
LIGHT?(2A)(EMIT? OR EMISSION?) OR EL OR E(W)L OR L(W)E(W)D
OR OLED OR LED

L30 8689 SEA SPE=ON ABB=ON PLU=ON L28 AND L29

L31 2171 SEA SPE=ON ABB=ON PLU=ON L30 AND PROC/RL

E VAPOR DEPOSITION PROCESS/CT

L32 246823 SEA SPE=ON ABB=ON PLU=ON "VAPOR DEPOSITION PROCESS"+PFT,
NT/CT

L33 251 SEA SPE=ON ABB=ON PLU=ON L31 AND L32

L34 12 SEA SPE=ON ABB=ON PLU=ON L33 AND (L21 OR L22 OR L23)

L35 414 SEA SPE=ON ABB=ON PLU=ON L30 AND (L21 OR L22 OR L23)

L36 89 SEA SPE=ON ABB=ON PLU=ON L35 AND L31

L37 12 SEA SPE=ON ABB=ON PLU=ON L36 AND CPS/RL

L38 11 SEA SPE=ON ABB=ON PLU=ON L36 AND (VAPOR DEPOSIT? OR
VAPOUR DEPOSIT?)

L39 23 SEA SPE=ON ABB=ON PLU=ON L34 OR (L37 OR L38)

L40 QUE SPE=ON ABB=ON PLU=ON REACTOR# OR (REACTION#) (2A)
(VESSEL# OR CHAMBER# OR TANK# OR SYSTEM# OR SPACE# OR
COMPARTMENT# OR RECEPTACLE# OR PORTION# OR PORT# OR
ASSEMBLY# OR SUB# (W) ASSEMBLY#)

L41 1 SEA SPE=ON ABB=ON PLU=ON L36 AND L40

L42 7 SEA SPE=ON ABB=ON PLU=ON L31 AND L40

L43 29 SEA SPE=ON ABB=ON PLU=ON L39 OR L41 OR L42

L44 20 SEA SPE=ON ABB=ON PLU=ON L43 AND (1840-2006)/PRY,AY,PY

L45 258 SEA SPE=ON ABB=ON PLU=ON L30 AND (L21 OR L22)

L46 13 SEA SPE=ON ABB=ON PLU=ON L45 AND L32

L47 3 SEA SPE=ON ABB=ON PLU=ON L45 AND L40

L48 71 SEA SPE=ON ABB=ON PLU=ON L45 AND (FEED? OR DELIVER? OR
SUPPLY? OR DISTRIBUT? OR TRANSPORT?)

L49 58 SEA SPE=ON ABB=ON PLU=ON L48 AND OPTIC?/SC,SX

L50 49 SEA SPE=ON ABB=ON PLU=ON L49 AND (1840-2006)/PRY,AY,PY

L51 65 SEA SPE=ON ABB=ON PLU=ON L44 OR L50

L52 5 SEA SPE=ON ABB=ON PLU=ON L51 AND L14

L53 QUE SPE=ON ABB=ON PLU=ON FILM? OR THINFILM? OR LAYER?
OR OVERLAY? OR OVERLAID? OR LAMIN? OR LAMEL? OR (MULTILAYER
?) OR SHEET? OR LEAF? OR FOIL? OR COAT? OR TOPCOAT? OR
OVERCOAT? OR VENEER? OR SHEATH? OR COVER? OR ENVELOP? OR
ENCAS? OR ENWRAP? OR OVERSPREAD?

L54 50 SEA SPE=ON ABB=ON PLU=ON L51 AND L53

L55 49 SEA SPE=ON ABB=ON PLU=ON L7 AND (L21 OR L22 OR L23)

L56 10 SEA SPE=ON ABB=ON PLU=ON L55 AND L29 AND L53

L57 21 SEA SPE=ON ABB=ON PLU=ON L55 AND L29

L58 21 SEA SPE=ON ABB=ON PLU=ON L56 OR L57

L59 18 SEA SPE=ON ABB=ON PLU=ON L58 AND (1840-2006)/PRY,AY,PY

L60 38 SEA SPE=ON ABB=ON PLU=ON L44 OR L59

L61 9 SEA SPE=ON ABB=ON PLU=ON L60 AND L50

L62 38 SEA SPE=ON ABB=ON PLU=ON (L60 OR L61)